Page 1

=> file reg

FILE 'REGISTRY' ENTERED AT 14:11:39 ON 12 SEP 2006 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2006 American Chemical Society (ACS)

Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 11 SEP 2006 HIGHEST RN 906423-10-7 DICTIONARY FILE UPDATES: 11 SEP 2006 HIGHEST RN 906423-10-7

New CAS Information Use Policies, enter HELP USAGETERMS for details.

TSCA INFORMATION NOW CURRENT THROUGH June 30, 2006

Please note that search-term pricing does apply when conducting SmartSELECT searches.

REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

http://www.cas.org/ONLINE/UG/regprops.html

=> file hcaplu FILE 'HCAPLUS' ENTERED AT 14:11:45 ON 12 SEP 2006 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2006 AMERICAN CHEMICAL SOCIETY (ACS)

Copyright of the articles to which records in this database refer is held by the publishers listed in the PUBLISHER (PB) field (available for records published or updated in Chemical Abstracts after December 26, 1996), unless otherwise indicated in the original publications. The CA Lexicon is the copyrighted intellectual property of the the American Chemical Society and is provided to assist you in searching databases on STN. Any dissemination, distribution, copying, or storing of this information, without the prior written consent of CAS, is strictly prohibited.

FILE COVERS 1907 - 12 Sep 2006 VOL 145 ISS 12 FILE LAST UPDATED: 11 Sep 2006 (20060911/ED)

New CAS Information Use Policies, enter HELP USAGETERMS for details.

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d que

STR / L2

Ak— C === 0 2 3

unsaturated carbofylic acid

NODE ATTRIBUTES:

CONNECT IS E1 RC AT CONNECT IS E3 RC AT WEINER 10/828468 09/12/2006 Page 2

DEFAULT MLEVEL IS ATOM GGCAT IS UNS AT 1 DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RSPEC I

NUMBER OF NODES IS

STEREO ATTRIBUTES: NONE L4SCR 2043

L6

STR Q

CH2: CH- G1 1 2 3

VAR G1=H/CH3

NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RSPEC I

NUMBER OF NODES IS

STEREO ATTRIBUTES: NONE L8 STR 3

O-CH2G1

CH2-CH2-O-C @4 5 6 7 1 2 3

VAR G1=4/9/12NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 14

STEREO ATTRIBUTES: NONE L10 STR 4

NODE ATTRIBUTES:

DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RSPEC I

NUMBER OF NODES IS

STEREO ATTRIBUTES: NONE

L12 112777 SEA FILE=REGISTRY SSS FUL L2 AND L6 AND (L8 OR L10) AND L4

L13 13901 SEA FILE=REGISTRY ABB=ON 74-85-1/CRN L14 6721 SEA FILE=REGISTRY ABB=ON 115-07-1/CRN

2359 SEA FILE=REGISTRY ABB=ON L12 AND (L13 OR L14) L15

ethylene or propylene

H3C--- CH2-O--- C CH2- O--- C 8 @9 10 11 @12 13 14

ethylene or propylene glycal

10

conductive composition, having an elastomer and/or a gum polymer dispersed in a matrix resin; where the elastomer and the gum polymer has an number average

particle size 0.05-5 μm . The separator is manufactured by mixing conductive particulates with a mixture of a premixed matrix resin and an elastomer and/or a gum polymer to obtain conductive composition; and molding the composition

The fuel cell has an **electrolyte** membrane between an electrode pair and the above separators stacked the **electrolyte**-electrode laminate.

IC ICM H01M008-02 ICS H01M008-10

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

IT 36704-47-9, Ethylene-glycidyl acrylate copolymer 106107-54-4,
Butadiene-styrene block copolymer

RL: DEV (Device component use); PRP (Properties); USES (Uses)

(compns. and manufacture of separators containing elastomers and/or rubbers with

controlled number average particle size for fuel cells)

IT 36704-47-9, Ethylene-glycidyl acrylate copolymer

RL: DEV (Device component use); PRP (Properties); USES (Uses)

(compns. and manufacture of separators containing elastomers and/or rubbers with

controlled number average particle size for fuel cells)

RN 36704-47-9 HCAPLUS

CN 2-Propenoic acid, oxiranylmethyl ester, polymer with ethene (9CI) (CA INDEX NAME)

CM 1

CRN 106-90-1 CMF C6 H8 O3

CM 2

CRN 74-85-1 CMF C2 H4

 $H_2C \longrightarrow CH_2$

L31 ANSWER 2 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2005:11686 HCAPLUS

DN 142:95239

TI Double bond-containing carbodimides and urethodiones, their derivatives and manufacture, **crosslinking** agents containing them, and their **crosslinked** polymers and applications

IN Aizawa, Wakana; Takada, Masakazu; Miura, Hidetoshi; Hyodo, Kenji; Ikegami, Koshiro; Fujita, Rei

PA Mitsubishi Paper Mills, Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 38 pp. CODEN: JKXXAF

DT Patent

LA Japanese

FAN	.CNT 1				
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	JP 2005002079	A2	20050106	JP 2003-193399	20030708 <
PRA:	I JP 2002-200322	A	20020709	<	
	JP 2002-304466	A	20021018	<	
	JP 2002-340421	Α	20021125	<	
	JP 2002-351679	A	20021203	<	
	JP 2002-368722	A	20021219	<	
	JP 2002-376484	Α	20021226	<	
	JP 2003-3105	Α	20030109	<	
	JP 2003-29004	Α	20030206	<	
	JP 2003-111573	Α	20030416	<	
os	MARPAT 142:95239				
GI					

Ι

The carbodiimides, urethodiones, and their derivs. are CH2:CR1CO2(CH2)lN:C:N(CH2)lCO2CR1:CH2 (I), II, and CH2:CR3CO2(CH2)nNHC[O(COQ)rCOCR3:CH2]:N(CH2)nCO2CR3:CH2 and III (R1-R4 = H, alkyl; Q = divalent linkage; l, m, n, p = 2-6; r = 0-5), resp. The polymers are useful for ion-conductive compns. for electrochem. devices, e.g., batteries, capacitors. Thus, Karenzu MOI (IV; 2-methacryloyloxyethyl isocyanate) was carbodiimized in the presence of p-nitrophenol as a thermal polymerization inhibitor and 3-methyl-1-phenyl-2-phospholene 1-oxide to give I (R1 = Me, l = 2), which was polymerized with NK Ester A 9300 and IV in nonaq. electrolytic solution comprising LiPF6, ethylene carbonate, and CO(OEt)2 to give a gel showing ion conductivity 4.9 + 10-3 S/cm at room temperature and no degradation after heating at 80° for 14 days. A secondary Li battery using the gel showed good durability.

IC ICM C07C267-00

ICS C07C275-70; C07D229-00; C08F020-36; H01B001-06; H01B001-12; H01G009-038; H01M006-18; H01M010-40; H01M014-00

CC 38-3 (Plastics Fabrication and Uses)

Section cross-reference(s): 23, 37, 52, 76

ST carbodiimide urethodione crosslinking agent manuf; ionic conductor crosslinked acrylic polycarbodiimide lithium salt; lithium battery electrolyte crosslinked acrylic polycarbodiimide; capacitor electrolyte crosslinked

polycarbodiimide; methacryloyloxyethyl isocyanate carbodiimidization nitrophenol thermal polymn inhibitor

IT Polycarbodiimides

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic, Li complexes, hexafluorophosphate-containing; manufacture of double

bond-containing carbodimides and urethodiones as **crosslinking** agents for **crosslinked** polymer gels as ionic conductors for electrochem. devices)

IT Polyoxyalkylenes, uses

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic, carbodiimide or urethodione derivs., Li complexes, hexafluorophosphate-containing; manufacture of double bond-containing carbodimides

and urethodiones as **crosslinking** agents for **crosslinked** polymer gels as ionic conductors for electrochem. devices)

IT Polyoxyalkylenes, preparation

RL: IMF (Industrial manufacture); PREP (Preparation)
(acrylic, carbodiimide or urethodione derivs.; manufacture of double bond-containing carbodimides and urethodiones as crosslinking agents for crosslinked polymer gels as ionic conductors for electrochem. devices)

IT Polyoxyalkylenes, uses

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic-polycarbodiimide-, Li complexes, hexafluorophosphate-containing; manufacture of double bond-containing carbodimides and urethodiones as crosslinking agents for crosslinked polymer gels as ionic conductors for electrochem. devices)

IT Polyoxyalkylenes, preparation

RL: IMF (Industrial manufacture); PREP (Preparation)
 (acrylic-polycarbodiimide-; manufacture of double bond-containing

carbodimides
 and urethodiones as crosslinking agents for
 crosslinked polymer gels as ionic conductors for electrochem.
 devices)

IT Polycarbodiimides

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic-polyoxyalkylene-, Li complexes, hexafluorophosphate-containing; manufacture of double bond-containing carbodimides and urethodiones as crosslinking agents for crosslinked polymer gels as ionic conductors for electrochem. devices)

IT Polycarbodiimides

RL: IMF (Industrial manufacture); PREP (Preparation)
(acrylic-polyoxyalkylene-; manufacture of double bond-containing carbodimides

and urethodiones as **crosslinking** agents for **crosslinked** polymer gels as ionic conductors for electrochem. devices)

IT Polycarbodiimides

RL: IMF (Industrial manufacture); PREP (Preparation)
(acrylic; manufacture of double bond-containing carbodimides and urethodiones as

crosslinking agents for crosslinked polymer gels as
ionic conductors for electrochem. devices)

IT Capacitors

```
(double layer; manufacture of double bond-containing carbodimides and
        urethodiones as crosslinking agents for crosslinked
        polymer gels as ionic conductors for electrochem. devices)
IT
     Secondary batteries
        (lithium; manufacture of double bond-containing carbodimides and
urethodiones as
        crosslinking agents for crosslinked polymer gels as
        ionic conductors for electrochem. devices)
IT
     Battery electrolytes
       Crosslinking agents
       Electrolytic capacitors
     Gels
     Ionic conductors
     Polymer electrolytes
     Secondary battery separators
        (manufacture of double bond-containing carbodimides and urethodiones as
        crosslinking agents for crosslinked polymer gels as
        ionic conductors for electrochem. devices)
IT
     Polyisocyanurates
     RL: DEV (Device component use); IMF (Industrial manufacture); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (polycarbodiimide-, acrylic, Li complexes, hexafluorophosphate-containing;
        manufacture of double bond-containing carbodimides and urethodiones as
        crosslinking agents for crosslinked polymer gels as
        ionic conductors for electrochem. devices)
IT
     Polyisocyanurates
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (polycarbodiimide-, acrylic; manufacture of double bond-containing
carbodimides
        and urethodiones as crosslinking agents for
        crosslinked polymer gels as ionic conductors for electrochem.
        devices)
IT
     Polycarbodiimides
     RL: DEV (Device component use); IMF (Industrial manufacture); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (polyisocyanurate-, acrylic, Li complexes, hexafluorophosphate-containing;
        manufacture of double bond-containing carbodimides and urethodiones as
        crosslinking agents for crosslinked polymer gels as
        ionic conductors for electrochem. devices)
TТ
     Polycarbodiimides
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (polyisocyanurate-, acrylic; manufacture of double bond-containing
carbodimides
        and urethodiones as crosslinking agents for
        crosslinked polymer gels as ionic conductors for electrochem.
        devices)
TΤ
     817619-88-8DP, tetraethylammonium complex, tetrafluoroborate-containing
     RL: DEV (Device component use); IMF (Industrial manufacture); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (double-layer capacitor electrolyte; manufacture of double
        bond-containing carbodimides and urethodiones as crosslinking
        agents for crosslinked polymer gels as ionic conductors for
        electrochem. devices)
IT
     817619-89-9P
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (film; manufacture of double bond-containing carbodimides and urethodiones
as
        crosslinking agents for crosslinked polymer gels as
```

ionic conductors for electrochem. devices)

```
101-68-8DP, carbodiimide or urethodione with dimethyl(isopropenyl)benzyl
     isocyanate homopolymer 822-06-0DP, Hexamethylene diisocyanate,
     carbodiimide or urethodione with methacryloyloxyethyl isocyanate
     homopolymer
                   88007-27-6DP, carbodiimide or urethodione derivative
     95627-99-9DP, carbodiimide or urethodione derivative
                                                            817619-73-1P
     817619-74-2P
                    817619-75-3P
                                   817619-77-5P
                                                  817619-78-6P
                                                                 817619-79-7P
     817619-80-0P
                    817619-81-1P
                                   817619-82-2P
                                                  817619-83-3DP, carbodiimide
     or urethodione derivative
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (manufacture of double bond-containing carbodimides and urethodiones as
        crosslinking agents for crosslinked polymer gels as
        ionic conductors for electrochem. devices)
IT
                    817619-69-5P
                                   817619-71-9P
                                                  817619-72-0P
     817619-67-3P
     RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
     (Reactant or reagent)
        (manufacture of double bond-containing carbodimides and urethodiones as
        crosslinking agents for crosslinked polymer gels as
        ionic conductors for electrochem. devices)
     79-41-4, Methacrylic acid, reactions
IT
                                            30674-80-7
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (manufacture of double bond-containing carbodimides and urethodiones as
        crosslinking agents for crosslinked polymer gels as
        ionic conductors for electrochem. devices)
     7439-93-2DP, Lithium, complex with carbodimide-containing polymers,
TΥ
     hexafluorophosphate-containing
                                      817619-74-2DP, Li complex,
     hexafluorophosphate-containing
                                     817619-79-7DP, Li complex,
     hexafluorophosphate-containing 817619-80-0DP, Li complex,
     hexafluorophosphate-containing
                                      817619-81-1DP, Li complex,
     hexafluorophosphate-containing
                                      817619-82-2DP, Li complex,
     hexafluorophosphate-containing
                                      817619-84-4DP, Li complex,
     hexafluorophosphate-containing
                                      817619-85-5DP, Li complex,
     hexafluorophosphate-containing
                                      817619-86-6DP, carbodiimide or urethodione
     derivative, Li complex, hexafluorophosphate-containing
                                                              817619-87-7DP, Li
     complex, hexafluorophosphate-containing
     RL: DEV (Device component use); IMF (Industrial manufacture); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (secondary Li battery electrolyte; manufacture of double
        bond-containing carbodimides and urethodiones as crosslinking
        agents for crosslinked polymer gels as ionic conductors for
        electrochem. devices)
TT
     817619-90-2DP, Li complex, hexafluorophosphate-containing
     RL: DEV (Device component use); IMF (Industrial manufacture); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (secondary Li battery separator and electrolyte; manufacture of
        double bond-containing carbodimides and urethodiones as
        crosslinking agents for crosslinked polymer gels as
        ionic conductors for electrochem. devices)
IT
     100-02-7, p-Nitrophenol, uses
     RL: CAT (Catalyst use); USES (Uses)
        (thermal polymerization inhibitor for carbodiimidization of isocyanates;
        manufacture of double bond-containing carbodimides and urethodiones as
        crosslinking agents for crosslinked polymer gels as
        ionic conductors for electrochem. devices)
IT
     817619-91-3P
                   817619-92-4P
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (water dispersion; manufacture of double bond-containing carbodimides and
        urethodiones as crosslinking agents for crosslinked
        polymer gels as ionic conductors for electrochem. devices)
     817619-90-2DP, Li complex, hexafluorophosphate-containing
IT
     RL: DEV (Device component use); IMF (Industrial manufacture); TEM
```

(Technical or engineered material use); PREP (Preparation); USES (Uses) (secondary Li battery separator and electrolyte; manufacture of double bond-containing carbodimides and urethodiones as crosslinking agents for crosslinked polymer gels as ionic conductors for electrochem. devices)

RN 817619-90-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, methanetetraylbis(nitrilo-2,1-ethanediyl) ester, polymer with ethene, α-(2-methyl-1-oxo-2-propenyl)-ω-methoxypoly(oxy-1,2-ethanediyl), α-(1-oxo-2-propenyl)-ω-methoxypoly(oxy-1,2-ethanediyl) and (2,4,6-trioxo-1,3,5-triazine-1,3,5(2H,4H,6H)-triyl)tri-2,1-ethanediyl tri-2-propenoate, graft (9CI) (CA INDEX NAME)

CM 1

CRN 817619-67-3 CMF C13 H18 N2 O4

CM 2

CRN 40220-08-4 CMF C18 H21 N3 O9

$$H_2C = CH - C - O - CH_2 - CH_2$$
 $CH_2 - CH_2 - O - C - CH = CH_2$
 $CH_2 - CH_2 - O - C - CH = CH_2$
 $CH_2 - CH_2 - O - C - CH = CH_2$

CM 3

CRN 32171-39-4

CMF (C2 H4 O)n C4 H6 O2

CCI PMS

$$H_2C = CH - C - CH_2 - CH_2 - CH_2 - OMe$$

CM 4

CRN 26915-72-0

CMF (C2 H4 O)n C5 H8 O2 CCI PMS

$$\begin{array}{c|c} {\rm H_2C} & {\rm O} & {\rm .} \\ & \parallel & \parallel & {\rm C} \\ {\rm Me-C-C-C-O-CH_2-CH_2-OMe} \end{array}$$
 OMe

CM 5

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

```
L31 ANSWER 3 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN
```

AN 2004:470551 HCAPLUS

DN 141:24797

TI Delamination- and heat-resistant laminates, their manufacture, and packaging materials using them

IN Suzuta, Masayoshi; Kurosawa, Akio

PA Toppan Printing Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 16 pp. CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

TIME. CIVI I						
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
PI JP 2004160766	A2	(200406)10	JP 2002-327847	20021112 <		
PRAT JP 2002-327847		200211/12	/			

AB The laminates, useful for packaging of Li batteries, etc., are manufactured by extrusion of reactive group-grafted thermoplastic resins, lamination on base materials, and hot rolling at temperature ≥10° higher than the m.p. of the resins. Thus, a boehmite-treated Al film was laminated with a blend of methacryloyloxyethyl isocyanate-grafted ethylene-1-octene copolymer and ethylene-1-hexene copolymer, dry-laminated with a polyamide, and made into a pouch showing good heat-seal strength and lamination strength even after storage with an electrolyte solution or a turf fungicide at 40° for 4 wk.

IC ICM B29C047-88

ICS B29C047-02; B32B015-08; B32B027-32; B65D065-40; C08F255-00; H01M002-02; B65D030-02; B29L009-00

CC 38-3 (Plastics Fabrication and Uses)
Section cross-reference(s): 5, 52

ST laminate packaging delamination resistant graft thermoplastic; battery electrolyte packaging laminate polyolefin graft; turf fungicide packaging laminate ethylene copolymer

IT Battery electrolytes

Laminated materials

(packaging; delamination - and heat-resistant laminates for packaging materials)

IT 115-07-1DP, Propylene, block polymers, grafted with functional group-containing methacrylates 189679-96-7P, Ethylene-glycidyl

methacrylate-1-octene graft copolymer 220772-59-8P, Ethylenemethacryloyloxyethyl isocyanate-1-octene graft copolymer 500785-61-5P,
Ethylene-methacryloyloxypropyltrimethoxysilane-1-octene graft copolymer
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM
(Technical or engineered material use); PREP (Preparation); USES (Uses)
(delamination- and heat-resistant laminates for packaging materials)
189679-96-7P, Ethylene-glycidyl methacrylate-1-octene graft

copolymer
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM
(Technical or engineered material use); PREP (Preparation); USES (Uses)
 (delamination- and heat-resistant laminates for packaging materials)

RN 189679-96-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester, polymer with ethene and 1-octene, graft (9CI) (CA INDEX NAME)

CM 1

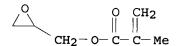
IT

CRN 111-66-0 CMF C8 H16

 $H_2C = CH - (CH_2)_5 - Me$

CM 2

CRN 106-91-2 CMF C7 H10 O3



CM 3

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

L31 ANSWER 4 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2004:330299 HCAPLUS

DN 140:340424

TI Manufacture of polyolefins containing less carboxylic acid residues for polymer electrolytes

IN Iwase, Yoshiyuki; Nishijima, Koichi; Ogasawara, Hiroshi; Kutsuwa, Yoshikazu

PA Du Pont-Mitsui Polychemicals Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 14 pp. CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

```
WEINER 10/828468 09/12/2006 -
                                 Page 12
     PATENT NO.
                         KIND
                                DATE
                                           APPLICATION NO.
                                                                   DATE
                         ____
                                           -----
ΡI
     JP 2004123872
                         A2
                               20040422
                                           JP 2002-289016
                                                                   20021001 <--
PRAI JP 2002-289016
                                20021001 <--
AB
     In the process, ethylene-unsatd. carboxylic acid copolymers are esterified
     with monohydroxy-terminated polyalkylene oxides and then reacted at
     residual carboxylic acids with end-capping agents to afford the claimed
     polyolefins useful for gel-type polymer batteries or capacitors. Thus,
     acrylic acid-ethylene copolymer (OH/carboxyl molar ratio 2.0) was
     esterified with polyethylene glycol monomethyl ether and then with benzoic
     acid to exhibit residual carboxylic acid 1.90% and high solubility in ethylene
     carbonate/propylene carbonate solvent after 6-mo storage at room temperature
IC
     ICM C08G081-02
     ICS H01B013-00; H01M010-40
     38-3 (Plastics Fabrication and Uses)
CC
     Section cross-reference(s): 52, 76
ST
     esterified endcapped residual carboxylic polyolefin electrolyte;
     durable polymer electrolyte residual acid minimized
TT
     Polyoxyalkylenes, uses
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (acrylic, graft, lower alkyl esters; manufacture of polyolefins containing
less
        carboxylic acid residues for polymer electrolytes)
IT
     Capacitors
        (electrolytes for; manufacture of polyolefins containing less
        carboxylic acid residues for polymer electrolytes)
IT
     Battery electrolytes
     Polymer electrolytes
        (manufacture of polyolefins containing less carboxylic acid residues for
polymer
        electrolytes)
IT
     103-71-9, Phenyl isocyanate, reactions
                                              111-26-2, n-Hexylamine
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (amidation agents; manufacture of polyolefins containing less carboxylic
acid
        residues for polymer electrolytes)
     680624-10-6DP, butylated 680972-65-0P, Acrylic
TΤ
     acid-ethylene-Uniox M 550 graft copolymer benzoate 680972-66-1P,
     Acrylic acid-ethylene-oxirane graft copolymer methyl ether benzoate
     680972-67-2DP, Acrylic acid-ethylene-oxirane graft copolymer
     methyl ether sodium salt, butylated
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (manufacture of polyolefins containing less carboxylic acid residues for
polymer
        electrolytes)
     680624-10-6DP, butylated 680972-65-0P, Acrylic
     acid-ethylene-Uniox M 550 graft copolymer benzoate 680972-66-1P,
     Acrylic acid-ethylene-oxirane graft copolymer methyl ether benzoate
     680972-67-2DP, Acrylic acid-ethylene-oxirane graft copolymer
     methyl ether sodium salt, butylated
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (manufacture of polyolefins containing less carboxylic acid residues for
polymer
        electrolytes)
     680624-10-6 HCAPLUS
RN
CN
     2-Propenoic acid, polymer with ethene and \alpha-methyl-\omega-
     hydroxypoly(oxy-1,2-ethanediyl), graft, sodium salt (9CI) (CA INDEX NAME)
```

CRN 680624-09-3

CMF (C3 H4 O2 . (C2 H4 O)n C H4 O . C2 H4)x

CCI PMS

CM 2

CRN 9004-74-4

CMF (C2 H4 O)n C H4 O

CCI PMS

CM 3

CRN 79-10-7 CMF C3 H4 O2

CM 4

CRN 74-85-1 CMF C2 H4

 $\mathtt{H}_2\mathtt{C} \underline{=\!=\!} \mathtt{C}\mathtt{H}_2$

RN 680972-65-0 HCAPLUS

CN 2-Propenoic acid, polymer with ethene and α -methyl- ω -hydroxypoly(oxy-1,2-ethanediyl), benzoate, graft (9CI) (CA INDEX NAME)

CM 1

CRN 65-85-0 CMF C7 H6 O2

C- OH

WEINER 10/828468 09/12/2006 · Page 14

CM 2

CRN 680624-09-3

CMF (C3 H4 O2 . (C2 H4 O) n C H4 O . C2 H4) x

CCI PMS

CM 3

CRN 9004-74-4

CMF (C2 H4 O)n C H4 O

CCI PMS

$$HO \longrightarrow CH_2 - CH_2 - O \longrightarrow n$$
 CH_3

CM 4

CRN 79-10-7 CMF C3 H4 O2

CM 5

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 680972-66-1 HCAPLUS

CN 2-Propenoic acid, polymer with ethene and oxirane, benzoate, methyl ether, graft (9CI) (CA INDEX NAME)

CM 1

CRN 67-56-1 CMF C H4 O

 $_{
m H_3C-OH}$

CM 2

CRN 65-85-0 CMF C7 H6 O2

CRN 196413-49-7

CMF (C3 H4 O2 . C2 H4 O . C2 H4) x

CCI PMS

CM 4

CRN 79-10-7 CMF C3 H4 O2

CM 5

CRN 75-21-8 CMF C2 H4 O

/0

CM 6

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 680972-67-2 HCAPLUS

CN 2-Propenoic acid, polymer with ethene and oxirane, methyl ether, graft, sodium salt (9CI) (CA INDEX NAME)

CM 1

CRN 67-56-1 CMF C H4 O

H₃C-OH

CRN 196413-49-7

CMF (C3 H4 O2 . C2 H4 O . C2 H4)x

· Page 16

CCI PMS

CM 3

CRN 79-10-7 CMF C3 H4 O2

О || но- с- сн— сн₂

CM 4

CRN 75-21-8 CMF C2 H4 O

/0

CM 5

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

L31 ANSWER 5 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2004:159965 HCAPLUS

DN 140:184769

TI Water vapor-barrier battery cases for sealed secondary batteries

IN Moritomi, Satoru; Yamaguchi, Takazo

PA Sumitomo Chemical Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 19 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 2004063302 A2 20040226 JP 2002-220868 20020730 <--

PRAI JP 2002-220868 20020730 <--

AB The cases have layers containing liquid crystalline polymers showing optical anisotropy during melting. The liquid crystalline polymers may be compns. having

continuous phases of liquid crystalline polyesters and dispersion phases of copolymers having functional groups (e.g., oxazolyl, epoxy, amino)

reactive to the polyesters. Preferably, the cases have laminates of the above layers and layers containing thermoplastic resins, e.g., polyolefins. The cases prevent evaporation of water in **electrolytic** solns. and prolong battery life.

IC ICM H01M002-02 ICS C08G081-00

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
Section cross-reference(s): 38

IT 51541-08-3P, Ethylene-glycidyl methacrylate-methyl acrylate copolymer

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (rubber, liquid crystalline layer containing; water vapor-barrier battery

cases

having liquid crystalline polymer layers for sealed secondary batteries)

IT 51541-08-3P, Ethylene-glycidyl methacrylate-methyl acrylate
copolymer

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (rubber, liquid crystalline layer containing; water vapor-barrier battery cases

having liquid crystalline polymer layers for sealed secondary batteries) RN 51541-08-3 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester, polymer with ethene and methyl 2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 106-91-2 CMF C7 H10 O3

CM 2

CRN 96-33-3 CMF C4 H6 O2

CM 3

CRN 74-85-1 CMF C2 H4

H2C=CH2

```
ANSWER 6 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN
L31
AN
     2003:719337 HCAPLUS
DN
     139:229682
TI
     Heat-sterilization of a packaged product and the heat-treated packaged
     product
     Yamane, Kazuyuki; Kawakami, Yukichika; Wakamatsu, Akiko; Yasuda, Matsuo;
IN
     Tanaka, Mikio
PA
     Kureha Chemical Industry Company, Limited, Japan
     PCT Int. Appl., 36 pp.
SO
     CODEN: PIXXD2
DT
     Patent
     English
LA
FAN.CNT 1
     PATENT NO.
                         KIND
                                DATE
                                           APPLICATION NO.
                                                                   DATE .
                                            ______
                         ----
                                -----
ΡI
     WO 2003074092
                         A1
                                20030912
                                            WO 2003-JP2431
                                                                   20030303 <--
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
             CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
             GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
             LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,
             PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ,
             UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,
             KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES,
             FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR,
             BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
     AU 2003210006
                                20030916
                                          AU 2003-210006
                          Α1
                                                                   20030303 <--
     EP 1480684
                                20041201
                                            EP 2003-743554
                          Α1
                                                                   20030303 <--
             AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
                                20050519
     US 2005106346
                                           US 2003-506571
                         Δ1
                                                                   20030303 <--
     JP 2005519824
                          T2
                                20050707
                                            JP 2003-579525
                                                                   20030303 <--
     CN 1638812
                                20050713
                                            CN 2003-804807
                          Α
                                                                   20030303 <--
PRAI JP 2002-57629
                          Α
                                20020304
                                         <--
     WO 2003-JP2431
                          W
                                20030303 <--
     A packaged product formed by enclosing a content material, such as food or
     beverage, sanitary products or medical products, within a packaging
     material including at least a layer of hydrophilic resin, is heat-treated
     with hot water for, e.g., boil sterilization or retort sterilization.
     hot water is caused to contain a water-soluble compound in an amount
     substantially exceeding a level contained in tap water, whereby it becomes
    possible to suppress the opalescence of the packaging material causing
     inferior appearance or transparency and leading to a lowering in
     gas-barrier property, which has been problematic in the conventional boil
    or retort hot water treatment.
IC
     ICM A61L002-04
         A23L003-02; A23L003-10; B65B055-10; A23B004-005; A23C003-023;
     ICS
         A61L002-26
     17-4 (Food and Feed Chemistry)
CC
     Section cross-reference(s): 62, 63
IT
    Beverages
    Cosmetics
    Drugs
      Electrolytes
     Food packaging materials
    Food preservation
    Food processing
    Packaging materials
    Sterilization and Disinfection
        (heat-sterilization of a packaged product and the heat-treated packaged
```

product)

IT 9003-07-0, Novatec FY6C 9010-79-1, Propylene-ethylene copolymer 24937-78-8, Evaflex V-527-4 24993-04-2, Amilan CM6001XF 25035-04-5, Rilsan Besvoafda 25067-34-9, Soarnol E 3803B 26061-90-5, 26124-68-5, Polyglycolic acid 26221-73-8, Moretec V Rexpearl Ra3150 0398CN 135373-05-6, Admer VF500 220181-18-0, Admer QB550 236738-49-1, Admer NF528 595570-75-5 595580-02-2, Supernyl SPR 8H 595580-08-8, Pairflex Sheet FA 292N RL: COS (Cosmetic use); FFD (Food or feed use); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (heat-sterilization of a packaged product and the heat-treated packaged

product)

IT 26061-90-5, Rexpearl Ra3150

RL: COS (Cosmetic use); FFD (Food or feed use); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

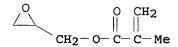
(heat-sterilization of a packaged product and the heat-treated packaged product)

RN 26061-90-5 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester, polymer with ethene (9CI) (CA INDEX NAME)

CM 1

CRN 106-91-2 CMF C7 H10 O3



CM 2

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L31 ANSWER 7 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2003:335197 HCAPLUS

DN 138:339113

TI Polyphenylene sulfide resin composition with good toughness and weld strength

IN Akiyama, Yoshikuni; Sakata, Minoru; Minonishi, Kuniaki

PA Asahi Kasei Kabushiki Kaisha, Japan; Asahi Kasei Chemicals Corporation

SO PCT Int. Appl., 31 pp. CODEN: PIXXD2

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE
PI WO 2003035760 A2 20030501 WO 2001-JP9399 20011025 <--

```
WO 2003035760
                          А3
                                20041223
            AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
             CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
             GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KR, KZ, LC, LK, LR, LS,
             LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT,
             RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US,
             UZ, VN, YU, ZA, ZW
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AM, AZ, BY, KG,
             KZ, MD, RU, TJ, TM, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR,
             IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN,
             GQ, GW, ML, MR, NE, SN, TD, TG
                          A2
                                20050223
                                            EP 2001-978917
                                                                    20011025 <--
            AT, BE, DE, FR
     CN 1620480
                                20050525
                                            CN 2001-823737
                                                                    20011025 <--
                          Α
     US 2004266951
                                20041230
                                            US 2004-492930
                                                                    20040419 <--
                          A1
PRAI WO 2001-JP9399
                          W
                                20011025
                                          <--
     Title composition comprises (A) a polyphenylene sulfide resin having a specific
     oligomer amount and a specific functional group amount, (B) a noncryst.
     thermoplastic resin having a glass transition temperature ≥120°,
     (C) and a compatibilizing agent having a specific functional group amount
     Use of this polyphenylene sulfide resin composition as a case material for a
     secondary battery enables the initial electrolyte performance to
     last over long. Thus, a composition comprising polyphenylene sulfide with melt
     viscosity at 300° and 20 kgf/cm2 load 500 P, amount oligomer extract in
     methylene chloride 0.4%, and SX (S = sulfur and X = metal or H) group
     content 29 µmol/g 71, polyphenylene ether with glass transition temperature
     209° and reduced viscosity 0.54 29, styrene-glycidyl methacrylate
     copolymer with weight average mol. weight 110,000 3.5, and hydrogenated
     styrene-butadiene triblock copolymer 17.6 parts was kneaded and
     injection-molded to give a test piece with heat distortion temperature
     135°, tensile strength 53 MPa, Izod impact resistance 137 J/m, weld
     tensile strength 52 MPa, and weld strength restion ratio 98%.
IC
     ICM C08L081-02
     ICS H01M002-02
CC
     37-6 (Plastics Manufacture and Processing)
     Section cross-reference(s): 52
IT
     24938-67-8, 2,6-Xylenol homopolymer, sru
                                                25134-01-4, 2,6-Xylenol
     homopolymer 26061-90-5, Bondfast 2C
     RL: POF (Polymer in formulation); PRP (Properties); TEM (Technical or
     engineered material use); USES (Uses)
        (blend with polyphenylene sulfide; polyphenylene sulfide resin composition
        with good toughness and weld strength)
IT
     26061-90-5, Bondfast 2C
     RL: POF (Polymer in formulation); PRP (Properties); TEM (Technical or
     engineered material use); USES (Uses)
        (blend with polyphenylene sulfide; polyphenylene sulfide resin composition
        with good toughness and weld strength)
RN
     26061-90-5 HCAPLUS
CN
     2-Propenoic acid, 2-methyl-, oxiranylmethyl ester, polymer with ethene
           (CA INDEX NAME)
     CM
          1
     CRN
         106-91-2
     CMF C7 H10 O3
```

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

```
ANSWER 8 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN
```

ΔN 2003:317760 HCAPLUS

138:341090 DN

Polymer gel electrolyte composition and its manufacture TI

Maruyama, Kunio; Miyagawa, Shinji; Yamaguchi, Shuichiro; Koyama, Noboru IN

Shirouma Science Co., Ltd., Japan; Fuji Heavy Industries Ltd.; Chemipro PA Kasei Ltd.; Mitsui and Co., Ltd. applicanta

SO Jpn. Kokai Tokkyo Koho, 16 pp.

CODEN: JKXXAF

DTPatent

LA Japanese

FAN.CNT 1

	PA	CENT :	NO.			KIN	D :	DATE			APPL	ICAT	ION I	NO.		D	ATE	
							-									-		
ΡI	JP	2003	1238	42		A2		2003	0425	,	JP 2	001-	3223	19		2	0011	019 <
	WO	O 2003036656				A1	20030501			WO 2002-JP10746				20021016 <				
		W:	ΑE,	AG,	AL,	AM,	ΑT,	AU,	ΑZ,	BA,	BB,	BG,	BR,	BY,	ΒZ,	CA,	CH,	CN,
			CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	ES,	FI,	GB,	GD,	GE,	GH,
			GM,	HR,	HU,	ID,	IL,	IN,	IS,	KΕ,	KG,	KP,	KR,	ΚZ,	LC,	LK,	LR,	LS,
			LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NO,	NZ,	OM,	PH,	PL,
			PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	ТJ,	TM,	TN,	TR,	TT,	TZ,	UA,
			UG,	US,	UΖ,	VC,	VN,	YU,	ZA,	ZM,	ZW							
		RW:	GH,	GM,	KΕ,	LS,	MW,	ΜZ,	SD,	SL,	SZ,	ΤZ,	ŪĠ,	ZM,	ZW,	AM,	ΑZ,	BY,
			KG,	ΚZ,	MD,	RU,	TJ,	TM,	ΑT,	ΒE,	BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,
			FI,	FR,	GB,	GR,	ΙE,	IT,	LU,	MC,	NL,	PT,	SE,	SK,	TR,	BF,	ВJ,	CF,
			CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	ΝE,	SN,	TD,	TG			
	TW	5934	98	B B		В	20040621		TW 2002-91124118				20021018 <					
	US	2004	1976	52		A1	1 2004100		1007	US 2004-828468					20040419 <			
PRAI	JP	2001	-322	319		Α		2001	1019	<-	-							
	WO	2002	-JP1	0746		A1		2002	1016	<	-							

The electrolyte composition, useful for electrochem. devices, has a 3-dimensional crosslinked structure of a crosslinked polymer network matrix in a mixed nonaq. solvent electrolyte solution, and a non-crosslinked polymer contained in the matrix; where the non-crosslinked polymer contains an ethylene unit and/or an propylene unit, and an unsatd. carboxylic acid obtained by esterizing a carboxyl group with a polyalkylene glycol protected by a hydroxyl group at its one end. The electrolyte composition is manufactured by dissolving the non-crosslinked polymer in the mixed nonaq. solvent electrolyte solution, adding a crosslinkable monomer to the mixture; and polymerizing the monomer with the mixture IC ICM H01M010-40

ICS C08G081-02; C08L023-26; C08L101-02; H01B001-06; H01G009-025;

H01G009-032

- CC. 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
- ST battery polymer gel electrolyte compn manuf
- IT Battery electrolytes

Polymer electrolytes

(compns. and manufacture of polymer gel electrolytes for electrochem. devices)

IT 518044-75-2P, Acrylic acid-ethylene copolymer, ester with polyethylene glycol monomethyl ether, polymer with polyethylene glycol diacrylate 518044-77-4P, Ethylene-methacrylic acid copolymer, ester with ethylene glycol monoethyl ether, polymer with polyethylene glycol diacrylate 518044-79-6P, Acrylic acid-ethylene copolymer, ester with ethylene glycol monomethyl ether, polymer with N-methylol methacrylamide 518044-81-0P, Ethylene-methacrylic acid copolymer, ester with ethylene glycol monoethyl ether, polymer with 3-hydroxyethyl methacrylate 518044-82-1P, Acrylic acid-ethylene copolymer, ester with ethylene glycol monomethyl ether, polymer with glycidyl acrylate 518044-83-2P, Acrylic acid-ethylene copolymer, ester with ethylene glycol monomethyl ether, polymer with 4,4'-diphenyl diisocyanate 518044-84-3P, Acrylic acid-ethylene copolymer, ester with ethylene glycol monomethyl ether, polymer with triphenyl methane triisocyanate 518044-86-5P, Ethylene-mathacrylic acid-propylene copolymer, ester with ethylene glycol monomethyl ether, polymer with polyethylene glycol diacrylate RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(compns. and manufacture of polymer gel electrolytes for electrochem. devices)

TT 96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate 108-32-7, Propylene carbonate 111-46-6, Diethylene glycol, uses 616-38-6, Dimethyl carbonate 623-53-0, Methyl ethyl carbonate 14283-07-9, Lithium tetrafluoroborate 21324-40-3, Lithium hexafluorophosphate 518044-78-5, Acrylic acid-ethylene copolymer, ester with ethylene glycol monomethyl ether, polymer with 1,6-hexanediol dimethacrylate RL: TEM (Technical or engineered material use); USES (Uses)

(compns. and manufacture of polymer gel electrolytes for electrochem. devices)

IT 518044-75-2P, Acrylic acid-ethylene copolymer, ester with polyethylene glycol monomethyl ether, polymer with polyethylene glycol diacrylate 518044-77-4P, Ethylene-methacrylic acid copolymer, ester with ethylene glycol monoethyl ether, polymer with polyethylene glycol diacrylate 518044-79-6P, Acrylic acid-ethylene copolymer, ester with ethylene glycol monomethyl ether, polymer with N-methylol methacrylamide 518044-81-0P, Ethylene-methacrylic acid copolymer, ester with ethylene glycol monoethyl ether, polymer with 3-hydroxyethyl methacrylate 518044-82-1P, Acrylic acid-ethylene copolymer, ester with ethylene glycol monomethyl ether, polymer with glycidyl acrylate 518044-83-2P, Acrylic acid-ethylene copolymer, ester with ethylene glycol monomethyl ether, polymer with 4,4'-diphenyl diisocyanate 518044-84-3P, Acrylic acid-ethylene copolymer, ester with ethylene glycol monomethyl ether, polymer with triphenyl methane triisocyanate 518044-86-5P, Ethylene-mathacrylic acid-propylene copolymer, ester with ethylene glycol monomethyl ether, polymer with polyethylene glycol diacrylate RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(compns. and manufacture of polymer gel electrolytes for electrochem. devices)

RN 518044-75-2 HCAPLUS

CN 2-Propenoic acid, polymer with ethene, ester with α -methyl- ω -

hydroxypoly(oxy-1,2-ethanediyl), graft, polymer with α -(1-oxo-2-propenyl)- ω -[(1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 26570-48-9 CMF (C2 H4 O)n C6 H6 O3 CCI PMS

$$H_2C = CH - C - CH_2 - CH_2$$

CM 2

CRN 177569-35-6 CMF (C3 H4 O2 . C2 H4)x . x (C2 H4 O)n C H4 O

CM 3

CRN 9004-74-4

CMF (C2 H4 O)n C H4 O

CCI PMS

HO
$$CH_2$$
 CH_2 O CH_3

CM 4

CRN 9010-77-9

CMF (C3 H4 O2 . C2 H4)x

CCI PMS

CM 5

CRN 79-10-7 CMF C3 H4 O2

$$\begin{matrix} 0 \\ || \\ \text{HO-C-CH} = \text{CH}_2 \end{matrix}$$

CM 6

CRN 74-85-1 CMF C2 H4 $H_2C = CH_2$

RN 518044-77-4 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with ethene, ester with \$\alpha\$-methyl-\$\omega\$-hydroxypoly(oxy-1,2-ethanediyl), graft, polymer with \$\alpha\$-(1-oxo-2-propenyl)-\$\omega\$-[(1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 26570-48-9

CMF (C2 H4 O)n C6 H6 O3

CCI PMS

$$H_2C = CH - C - CH_2 - CH_2$$

CM 2

CRN 518044-76-3

CMF (C4 H6 O2 . C2 H4)x . x (C2 H4 O)n C H4 O

CM 3

CRN 9004-74-4

CMF (C2 H4 O)n C H4 O

CCI PMS

CM 4

CRN 25053-53-6

CMF (C4 H6 O2 . C2 H4)x

CCI PMS

CM 5

CRN 79-41-4 CMF C4 H6 O2

CH₂ || Me- C- CO₂H

CM 6

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 518044-79-6 HCAPLUS

CN 2-Propenoic acid, polymer with ethene, ester with α -methyl- ω -hydroxypoly(oxy-1,2-ethanediyl), graft, polymer with N-(hydroxymethyl)-2-methyl-2-propenamide (9CI) (CA INDEX NAME)

CM 1

CRN 923-02-4 CMF C5 H9 N O2

$$^{\mathrm{H_2C}}_{\parallel\parallel\parallel}$$
 O Me-C-C-NH-CH₂-OH

CM 2

CRN 177569-35-6 CMF (C3 H4 O2 . C2 H4)x . x (C2 H4 O)n C H4 O

CM 3

CRN 9004-74-4

CMF (C2 H4 O)n C H4 O

CCI PMS

CM 4

CRN 9010-77-9

CMF (C3 H4 O2 . C2 H4)x

CCI PMS

CM 5

CRN 79-10-7 CMF C3 H4 O2

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 518044-81-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with ethene, ester with α -methyl- ω -hydroxypoly(oxy-1,2-ethanediyl), graft, polymer with 2-hydroxyethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 868-77-9 CMF C6 H10 O3

CM 2

CRN 518044-76-3 CMF (C4 H6 O2 . C2 H4)x . x (C2 H4 O)n C H4 O

CM 3

CRN 9004-74-4

CMF (C2 H4 O)n C H4 O

CCI PMS

CM 4

CRN 25053-53-6

CMF (C4 H6 O2 . C2 H4)x

CCI PMS

CM 5

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me--C--CO}_2 \text{H} \end{array}$$

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 518044-82-1 HCAPLUS

CN 2-Propenoic acid, polymer with ethene, ester with α -methyl- ω -hydroxypoly(oxy-1,2-ethanediyl), graft, polymer with oxiranylmethyl 2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 106-90-1 CMF C6 H8 O3

CM 2

CRN 177569-35-6

CMF (C3 H4 O2 . C2 H4)x . x (C2 H4 O)n C H4 O

CM 3

CRN 9004-74-4

CMF (C2 H4 O)n C H4 O

CCI PMS

CM 4

CRN 9010-77-9

CMF (C3 H4 O2 . C2 H4) \times

CCI PMS

CM 5

CRN 79-10-7

CMF C3 H4 O2

CM 6

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 518044-83-2 HCAPLUS

CN 2-Propenoic acid, polymer with ethene, ester with α -methyl- ω -hydroxypoly(oxy-1,2-ethanediyl), graft, polymer with 1,1'-methylenebis[4-isocyanatobenzene] (9CI) (CA INDEX NAME)

CM 1

CRN 101-68-8 CMF C15 H10 N2 O2

CM 2

CRN 177569-35-6

CMF (C3 H4 O2 . C2 H4)x . x (C2 H4 O)n C H4 O

CM 3

CRN 9004-74-4

CMF (C2 H4 O)n C H4 O

CCI PMS

$$HO \longrightarrow \begin{bmatrix} CH_2 - CH_2 - O & \\ & n \end{bmatrix}$$
 CH₃

CM 4

CRN 9010-77-9

CMF (C3 H4 O2 . C2 H4)x

CCI PMS

WEINER 10/828468 09/12/2006 Page 29

CM 5

CRN 79-10-7 CMF C3 H4 O2

CM 6

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 518044-84-3 HCAPLUS

CN 2-Propenoic acid, polymer with ethene, ester with α -methyl- ω -hydroxypoly(oxy-1,2-ethanediyl), graft, polymer with 1,1',1''-methylidynetris[isocyanatobenzene] (9CI) (CA INDEX NAME)

CM 1

CRN 25656-78-4 CMF C22 H13 N3 O3 CCI IDS



D1-NCO

CM 2

CRN 177569-35-6 CMF (C3 H4 O2 . C2 H4)x . x (C2 H4 O)n C H4 O

CM 3

CRN 9004-74-4

CMF (C2 H4 O)n C H4 O

CCI PMS

$$HO \longrightarrow CH_2 - CH_2 - O \longrightarrow n$$
 CH_3

CRN 9010-77-9 CMF (C3 H4 O2 . C2 H4)x CCI PMS

CM 5

CRN 79-10-7 CMF C3 H4 O2

CM 6

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 518044-86-5 HCAPLUS

CM 1

CRN 26570-48-9 CMF (C2 H4 O)n C6 H6 O3

CCI PMS

$$H_2C = CH - C - CH_2 - CH_2$$

CM 2

CRN 518044-85-4 CMF (C4 H6 O2 . C3 H6 . C2 H4)x . x (C2 H4 O)n C H4 O

CRN 9004-74-4

CMF (C2 H4 O)n C H4 O

CCI PMS

CM 4

CRN 28433-68-3

CMF (C4 H6 O2 . C3 H6 . C2 H4)x

CCI PMS

CM 5

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

CM 6

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ \parallel \\ \text{Me-} \text{C-} \text{CO}_2 \text{H} \end{array}$$

CM 7

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 518044-78-5 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 1,6-hexanediyl ester, polymer with ethene graft polymer with 2-propenoic acid ester with α -methyl- ω -hydroxypoly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CRN 6606-59-3 CMF C14 H22 O4

CM 2

CRN 177569-35-6 CMF (C3 H4 O2 . C2 H4)x . x (C2 H4 O)n C H4 O

CM 3

CRN 9004-74-4 CMF (C2 H4 O)n C H4 O CCI PMS

CM 4

CRN 9010-77-9 CMF (C3 H4 O2 . C2 H4)x CCI PMS

CM 5

CRN 79-10-7 CMF C3 H4 O2

CM 6

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

L31 ANSWER 9 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN AN 2002:315264 HCAPLUS

```
DN
     136:343316
TI
     Gel-type polymer electrolyte that can be molded to a
     self-supported film for lithium batteries
     Oyama, Noboru; Fujimoto, Yuki; Iwase, Yoshiyuki; Nishijima, Kouichi
IN
PA
     Du Pont-Mitsui Polychemicals Co., Ltd., Japan
     PCT Int. Appl., 50 pp.
SO
     CODEN: PIXXD2
TG
     Patent
LA
     English
FAN.CNT 1
                                DATE/
     PATENT NO.
                         KIND
                                            APPLICATION NO.
                                                                    DATE
                         ----
                                            -----
                                                                    -----
                                20020425
                                            WO 2001-JP9138
     WO 2002033765
                          A2
                                                                    20011018 <--
PΙ
                                200/31002
     WO 2002033765
                          Α3
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
             CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
             GM, HR, HU, ID, IL, IN, IS, KE, KG, KP, KR, KZ, LC, LK, LR, LS,
             LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT,
             RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US,
             UZ, VN, YU, ZA, ZW
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AM, AZ, BY, KG,
             KZ, MD, RU, TJ, TM, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR,
             IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN,
             GQ, GW, ML, MR, NE, SN, TD, TG
     CA 2426129
                          AA
                                20020425
                                            CA 2001-2426129
                                                                    20011018 <--
     JP 2002198095
                         , A2
                                            JP 2001-320319
                                20020712
                                                                    20011018 <--
     EP 1368849
                                            EP 2001-976730
                                20031210
                                                                    20011018 <--
                          A2
     EP 1368849
                                20060405
                          В1
             AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
                                                                    20011018 <--
     CN 1555589
                          Α
                                20041215
                                          CN 2001-820726
PRAI JP 2000-318169
                          Α
                                20001018 <--
                                20011018 <--
     WO 2001-JP9138
                          W
     In a gel-type polymer electrolyte, the polymer comprises (a) an
AB
     ethylene-unsatd. carboxylic acid copolymer or a derivative thereof and (b) a
     polyalkylene oxide having a hydroxyl group at one terminal thereof or a
     derivative thereof, which are bonded together by an ester bond.
                                                                       The gel-type
     polymer electrolyte has a high ionic conductivity, and makes it
     possible to provide a cell which has excellent charge/discharge
     characteristics at low temps. as well as at high temps.
IC
     ICM HO1M
     52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
CC .
     Section cross-reference(s): 38, 76
ST
     lithium battery gel type polymer electrolyte
IT
     Battery electrolytes
     Capacitors
     Ionic conductivity
     Swelling, physical
     Transesterification
        (gel-type polymer electrolyte that can be molded to
        self-supported film for lithium batteries)
IT
     Polyoxyalkylenes, uses
     RL: DEV (Device component use); USES (Uses)
        (gel-type polymer electrolyte that can be molded to
        self-supported film for lithium batteries)
IT
     Secondary batteries
        (lithium; gel-type polymer electrolyte that can be molded to
        self-supported film for lithium batteries)
IT
     Alcohols, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
```

```
(polyhydric, crosslinking agent; gel-type polymer
        electrolyte that can be molded to self-supported film for
        lithium batteries)
     Fatty acids, reactions
IT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (unsatd., crosslinking agent; gel-type polymer
        electrolyte that can be molded to self-supported film for
        lithium batteries)
TΤ
     Fatty acids, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (unsatd., esters, crosslinking agent; gel-type polymer
        electrolyte that can be molded to self-supported film for
        lithium batteries)
IT
     79-41-4, Methacrylic acid, reactions
                                            18358-13-9, Methacrylate, reactions
     25721-76-0, Polyethylene glycol dimethacrylate 26403-72-5, Polyethylene
     glycol diglycidyl ether
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (crosslinking agent; gel-type polymer electrolyte
        that can be molded to self-supported film for lithium batteries)
TΤ
     96-48-0, γ-Butyrolactone 96-49-1, Ethylene carbonate
                                                              105-58-8,
                         108-32-7, Propylene carbonate
     Diethyl carbonate
                                                         110-71-4
                          872-50-4, n-Methylpyrrolidone, uses
     Dimethyl carbonate
                                                                 14283-07-9,
                                 21324-40-3, Lithium hexafluorophosphate
     Lithium tetrafluoroborate
     35895-69-3, Tetraethylammonium trifluoromethanesulfonate
     RL: DEV (Device component use); USES (Uses)
        (gel-type polymer electrolyte that can be molded to
        self-supported film for lithium batteries)
ΙT
     9004-74-4DP, Polyethylene glycol monomethyl ether, reaction product of
     acrylic acid-ethylene copolymer 172588-43-1DP, Ethylene glycol-propylene
     glycol mono-2-ethylhexyl ether block copolymer, reaction products with
     acrylic acid-ethylene copolymer 177569-35-6DP, reaction product
     polyethylene glycol monomethyl ether 177569-35-6DP, reaction
     products with acrylic acid-ethylene copolymer 196521-53-6DP,
     reaction products with acrylic acid-ethylene copolymer
     RL: DEV (Device component use); SPN (Synthetic preparation); PREP
     (Preparation); USES (Uses)
        (gel-type polymer electrolyte that can be molded to
        self-supported film for lithium batteries)
IT
     104-15-4, p-Toluenesulfonic acid, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (gel-type polymer electrolyte that can be molded to
        self-supported film for lithium batteries)
IT
     177569-35-6DP, reaction product polyethylene glycol monomethyl
     ether 196521-53-6DP, reaction products with acrylic
     acid-ethylene copolymer
     RL: DEV (Device component use); SPN (Synthetic preparation); PREP
     (Preparation); USES (Uses)
        (gel-type polymer electrolyte that can be molded to
        self-supported film for lithium batteries)
RN
     177569-35-6 HCAPLUS
CN
     2-Propenoic acid, polymer with ethene, ester with \alpha\text{-methyl-}\omega\text{-}
     hydroxypoly(oxy-1,2-ethanediyl), graft (9CI) (CA INDEX NAME)
     CM
     CRN
         9004-74-4
     CMF
          (C2 H4 O)n C H4 O
     CCI PMS
```

HO
$$CH_2$$
 CH_2 O CH_3

CRN 9010-77-9

CMF (C3 H4 O2 . C2 H4) \times

CCI PMS

CM 3

CRN 79-10-7 CMF C3 H4 O2

$$\begin{matrix} \text{O} \\ || \\ \text{HO-C-CH} = \text{CH}_2 \end{matrix}$$

CM 4

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 196521-53-6 HCAPLUS

CN 2-Propenoic acid, polymer with ethene and oxirane, methyl ether, graft (9CI) (CA INDEX NAME)

CM 1

CRN 67-56-1 CMF C H4 O

н3С-он

CM 2

CRN 196413-49-7

CMF (C3 H4 O2 . C2 H4 O . C2 H4) x

CCI PMS

CM 3

CRN 79-10-7 CMF C3 H4 O2 $\begin{matrix} \text{O} \\ || \\ \text{HO-C-CH} = \text{CH}_2 \end{matrix}$

CM 4

CRN 75-21-8 CMF C2 H4 O

 \triangle

CM 5

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

L31 ANSWER 10 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2001:545760 HCAPLUS

DN 135:137856

TI Olefin block copolymers with linking groups, production processes of the same and use thereof

IN Ota, Seiji; Moriya, Satoru; Mori, Ryoji; Koda, Taku; Tan, Junji; Kojoh, Shinichi; Kaneko, Hideyuki; Hama, Shunichi; Nobori, Tadahito; Matsugi, Tomoaki; Kashiwa, Norio

PA Mitsui Chemicals, Inc., Japan

SO PCT Int. Appl., 563 pp.

CODEN: PIXXD2

DT Patent

LA Japanese

FAN.CNT 4

ran.	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
ΡI	WO 2001053369	A1	20010726	WO 2001-JP298	20010118 <		
	W: CN, KR, SO RW: DE, FR, GE	•	,				
	JP 2002097237	A2	20020402	JP 2000-288181	20000922 <		
	EP 1275670	A1	20030115	EP 2001-942647	20010118 <		
	EP 1275670	В1	20050810				
	R: DE, FR, GE	3					
	JP 2001278928	A2	20011010	JP 2001-16069	20010124 <		
	JP 2001278929	A2	20011010	JP 2001-16070	20010124 <		
	JP 2001278930	A2	20011010	JP 2001-18285	20010126 <		
	JP 2001278931	A2	20011010	JP 2001-18299	20010126 <		
	JP 2001278932	A2	20011010	JP 2001-18300	20010126 <		
	JP 2001288443	A2	20011016	JP 2001-25809	20010201 <		
	JP 2001288372	A2	20011016	JP 2001-25810	20010201 <		
	JP 2001288272	A2	20011016	JP 2001-25811	20010201 <		
	JP 2001342256	A2	20011211	JP 2001-76944	20010316 <		

```
WEINER 10/828468
                   09/12/2006 -
                                    Page 37
     JP 2001348413
                          A2
                                20011218
                                             JP 2001-106007
                                                                    20010404 <--
     JP 2002012639
                          A2
                                20020115
                                             JP 2001-106006
                                                                    20010404 <--
                          A2
     JP 2002037825
                                20020206
                                            JP 2001-141562
                                                                    20010511 <--
                          A2
                                                                    20010511 <--
     JP 2002053632
                                20020219
                                            JP 2001-141561
     US 2003055179
                          A1
                                20030320
                                            US 2002-181553
                                                                    20020719 <--
PRAI JP 2000-17848
                          Α
                                20000121
                                          <--
     JP 2000-17849
                          Α
                                20000121 <--
     JP 2000-17850
                          Α
                                20000121 <--
     JP 2000-18053
                          Α
                                20000125 <--
     JP 2000-18054
                          Α
                                20000125
                                          <--
     JP 2000-23333
                          Α
                                20000127
                                          <--
     JP 2000-24736
                          Α
                                20000128
                                          <--
     JP 2000-24737
                          Α
                                20000128
                                          <--
     JP 2000-28924
                          Α
                                20000201 <--
     JP 2000-28925
                          Α
                                20000201 <--
     JP 2000-28926
                          Α
                                20000201 <--
     JP 2000-90716
                          Α
                                20000327
                                          <---
     JP 2000-111900
                          Α
                                20000407
                                          <--
                          Α
     JP 2000-132859
                                20000427
                                          <--
                          Α
     JP 2000-147500
                                20000515
                                          <--
     JP 2000-166470
                          Α
                                20000531 <--
     JP 2000-288181
                          Α
                                20000922 <--
     WO 2001-JP298
                          W
                                20010118 <--
     The olefin block copolymers have excellent in affinity with metal, polar
AB
     resins or the like, impact resistance, mar resistance, thermal resistance,
     rigidity, oil resistance, transparency, antifogging properties, elec.
     insulation properties, breakdown voltage, application properties,
     low-temperature flexibility, moldability, environmental degradation properties,
     fluidity and/or dispersion properties. The block copolymers are of
     PO1-g1-B1 type (wherein PO1 is a segment composed of repeating units
     derived from C2-20 olefin; g1 is an ester, ether, amide, imide, urethane,
     urea, silyl ether, or carbonyl linkage; and B1 is an unsatd. hydrocarbon
     or heteroatom-containing segment), and are useful for hot-melt adhesives,
     moldings, modifiers for plastics and rubbers, etc. Thus, a copolymer with
     O linking group was prepared, e.g., by metallocene polymerization of ethylene
with
     norbornene, followed by converting the resulting single-end unsatd.
     group-terminated copolymer to a B-terminated copolymer using 9-BBN
     (9-borabicyclononane), and block copolymn. in THF with styrene in the
     presence of O.
TC
     C08F293-00; C08G081-00; C08L053-00; C08L101-00
CC
     35-4 (Chemistry of Synthetic High Polymers)
     Section cross-reference(s): 39
IT
     Boronizing
     Bottles
     Coating materials
     Containers
     Inks
     Oxidation
     Pipes and Tubes
     Plastic films
     Polymer electrolytes
     Sealing compositions
     Transparent materials
        (polymer end conversion in manufacture of olefin diblock copolymers for use
        in plastics and rubbers)
TT
     350846-86-5P
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
```

HO-CH2-CH2-O-C-CH-CH2

```
(crosslinked; olefinic polymer end conversion in manufacture of
        block copolymers and various uses in plastics and rubbers)
     126140-91-8P, Ethylene-ethylene oxide block copolymer
IT
     Maleic anhydride-propylene-styrene block copolymer
                                                         188448-09-1P,
     Ethylene-methyl methacrylate-propylene block copolymer
                                                              350846-74-1P
     350846-75-2P
                    350846-76-3P 350846-77-4P
                                                350846-78-5P
     351471-95-9DP, amine-terminated
                                       718637-95-7P
                                                      729589-59-7P
     733035-84-2DP, amine-terminated
                                       733036-88-9P
                    733037-64-4P
     733036-90-3P
                                   733037-66-6P
                                                 733037-80-4P
     737795-80-1P
     RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (diblock; olefinic polymer end conversion in manufacture of block copolymers
        and various uses in plastics and rubbers)
IT
     110341-23-6P, Methyl methacrylate-propylene block copolymer
     141551-66-8P, ε-Caprolactone-ethylene block copolymer
     185630-55-1P, Ethyl methacrylate-propylene block copolymer
                                                                  185630-56-2P,
     Butyl acrylate-propylene block copolymer 330665-83-3P, Butyl
     acrylate-ethylene block copolymer 350846-83-2P
                                                     350846-85-4P
                                  350846-89-8P
     350846-87-6P
                    350846-88-7P
                                                 350846-90-1P
                                                                 350846-91-2P
     351471-97-1P, Ethylene-methyl methacrylate-1-octene block copolymer
     351471-98-2P
                   351471-99-3P
                                  351472-00-9P 351472-01-0P
                    731774-35-9P 731774-40-6P
     351472-02-1P
                                                731842-58-3P
                                   733035-76-2P
     733035-24-0P
                   733035-27-3P
                                                  733035-86-4P
                                                                 733035-87-5P
     733035-93-3P
                                   733036-06-1P
                   733035-96-6P
                                                  733036-15-2P
     733036-85-6P
                    733038-31-8P
                                   736998-02-0P
                                                  736998-15-5P
     743421-29-6P
                    790684-53-6P
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (diblock; olefinic polymer end conversion in manufacture of block copolymers
        and various uses in plastics and rubbers)
IT
     350846-82-1P
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (triblock, star, 4-arm; olefinic polymer end conversion in manufacture of
        block copolymers and various uses in plastics and rubbers)
IT
     350846-77-4P 351471-95-9DP, amine-terminated
     733035-84-2DP, amine-terminated 733036-90-3P
     RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation): USES (Uses)
        (diblock; olefinic polymer end conversion in manufacture of block copolymers
        and various uses in plastics and rubbers)
RN
     350846-77-4 HCAPLUS
     2-Propenoic acid, 2-hydroxyethyl ester, polymer with ethene and 1-hexene,
CN
                (CA INDEX NAME)
     block (9CI)
     CM
          1
     CRN
        818-61-1
     CMF C5 H8 O3
```

CRN 592-41-6 CMF C6 H12

 $H_2C = CH - Bu - n$

CM 3

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 351471-95-9 HCAPLUS

CN 2-Propenoic acid, 2-hydroxyethyl ester, polymer with ethene and 1-octene, block (9CI) (CA INDEX NAME)

CM 1

CRN 818-61-1 CMF C5 H8 O3

 $\begin{array}{c} & \text{O} \\ || \\ \text{HO-CH}_2\text{-CH}_2\text{-O-C-CH-----} \text{CH}_2 \end{array}$

CM 2

CRN 111-66-0 CMF C8 H16

 $H_2C = CH - (CH_2)_5 - Me$

CM 3

CRN 74-85-1 CMF C2 H4

 $H_2C \longrightarrow CH_2$

RN 733035-84-2 HCAPLUS

CN 2-Propenoic acid, 2-hydroxyethyl ester, polymer with ethene and 1-octene, diblock (9CI) (CA INDEX NAME)

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} & \circ \\ || \\ \text{HO-CH}_2\text{-CH}_2\text{-O-C-CH-----} \text{CH}_2 \end{array}$$

CM2

CRN 111-66-0 CMF C8 H16

 $H_2C = CH - (CH_2)_5 - Me$

CM 3

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

733036-90-3 HCAPLUS RN

2-Propenoic acid, 2-hydroxyethyl ester, polymer with ethene and 1-hexene, CNdiblock (9CI) (CA INDEX NAME)

CM 1

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} {\rm O} \\ || \\ {\rm HO-CH_2-CH_2-O-C-CH} \end{array}$$

CM 2

CRN 592-41-6 CMF C6 H12

 $H_2C = CH - Bu - n$

CM 3

CRN 74-85-1

CMF C2 H4

 $H_2C = CH_2$

IT 350846-83-2P 351472-01-0P 731774-40-6P

733036-85-6P

RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(diblock; olefinic polymer end conversion in manufacture of block copolymers and various uses in plastics and rubbers)

RN 350846-83-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-methoxyethyl ester, polymer with methyl 2-methyl-2-propenoate and 1-propene, block (9CI) (CA INDEX NAME)

CM 1

CRN 6976-93-8 CMF C7 H12 O3

 $\begin{array}{c|c} ^{\rm H_2C} & {\rm O} \\ & \parallel & \parallel \\ {\rm Me^-\,C^-\,C^-\,O^-\,CH_2^-\,CH_2^-\,OMe} \end{array}$

CM 2.

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

CM 3

CRN 80-62-6 CMF C5 H8 O2

 $\begin{array}{c|c} ^{H_2\mathbb{C}} & \text{O} \\ \parallel & \parallel \\ \text{Me-} \text{C-} \text{C-} \text{OMe} \end{array}$

RN 351472-01-0 HCAPLUS

CN 2-Propenoic acid, 2-hydroxyethyl ester, polymer with 1-propene, block (9CI) (CA INDEX NAME)

CM 1

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} {\rm O} \\ || \\ || \\ {\rm HO-CH_2-CH_2-O-C-CH} \end{array} \\ {\rm CH_2} \\$$

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

RN 731774-40-6 HCAPLUS

CN 2-Propenoic acid, 2-hydroxyethyl ester, polymer with 1-propene, diblock (9CI) (CA INDEX NAME)

CM 1

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} {\rm O} \\ || \\ {\rm HO-CH_2-CH_2-O-C-CH-} \end{array}$$

CM 2

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

RN 733036-85-6 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-methoxyethyl ester, polymer with methyl 2-methyl-2-propenoate and 1-propene, diblock (9CI) (CA INDEX NAME)

CM 1

CRN 6976-93-8 CMF C7 H12 O3

CM 2

CRN 115-07-1 CMF C3 H6 $H_3C-CH=CH_2$

CM 3

CRN 80-62-6 CMF C5 H8 O2

 $\begin{array}{c|c} ^{H_2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-} \text{C-} \text{C-} \text{OMe} \end{array}$

IT 350846-82-1P

RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(triblock, star, 4-arm; olefinic polymer end conversion in manufacture of block copolymers and various uses in plastics and rubbers)

RN 350846-82-1 HCAPLUS

CN 2-Propenoic acid, butyl ester, polymer with ethene, oxirane and 1-propene, block (9CI) (CA INDEX NAME)

CM 1

CRN 141-32-2 CMF C7 H12 O2

CM 2

CRN 115-07-1 CMF C3 H6

 $H_3C-CH-CH_2$

CM 3

CRN 75-21-8 CMF C2 H4 O

 $\stackrel{\circ}{\triangle}$

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L31 ANSWER 11 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2000:418095 HCAPLUS

DN 133:65226

TI Ion-conducting polyoxyalkylene-based **electrolyte** containing boron complex lithium salt

IN Takeda, Shoichi; Shindo, Nobumitsu

PA Japan Carlit Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 10 pp. CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

FAN.CNI I						
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
PI JP 2000173343	A2	20000623	JP 1998-343040	19981202 <		
PRAI JP 1998-343040		19981202 <	; - -			
OS MARPAT 133:65226						
GI						

$$\begin{bmatrix} R^1 & 0 & 0 & R^2 \\ R^2 & 0 & 0 & R^1 \end{bmatrix}$$
 lit

- AB The ion conductor **electrolyte** consists of a B complex Li salt I [R1 = H, C1-4 alkyl, (substituted) benzene ring, (substituted) naphthalene ring; R2 = (substituted) benzene ring, (substituted) naphthalene ring] and a polymer involving polyoxyalkylene segments. The **electrolyte** shows improved corrosion inhibition and heat resistance.
- IC ICM H01B001-06 ICS C08K005-55; C08L055-00; C08L071-02; C07F005-02; C08F290-06; H01M006-18; H01M010-40

Ι

CC 72-3 (Electrochemistry)

Section cross-reference(s): 38, 52, 76

- ST ion conducting polyoxyalkylene **electrolyte** corrosion inhibition; boron complex lithium salt doped polymer; heat resistance polyoxyalkylene **electrolyte**
- IT Polyoxyalkylenes, uses
 - RL: TEM (Technical or engineered material use); USES (Uses) (acrylic; polyoxyalkylene containing boron complex lithium salt as ion-conducting electrolyte with improved heat resistance and corrosion inhibition)

IT Corrosion prevention

Heat-resistant materials

Polymer electrolytes

(polyoxyalkylene containing boron complex lithium salt as ion-conducting electrolyte with improved heat resistance and corrosion inhibition)

IT Coordination compounds

RL: MOA (Modifier or additive use); USES (Uses) (polyoxyalkylene containing boron complex lithium salt as ion-conducting electrolyte with improved heat resistance and corrosion inhibition)

IT Polyoxyalkylenes, uses

RL: TEM (Technical or engineered material use); USES (Uses)
(polyoxyalkylene containing boron complex lithium salt as ion-conducting
electrolyte with improved heat resistance and corrosion
inhibition)

IT 134587-71-6 166097-73-0 273929-21-8 276889-02-2

RL: MOA (Modifier or additive use); USES (Uses)

(polyoxyalkylene containing boron complex lithium salt as ion-conducting electrolyte with improved heat resistance and corrosion inhibition)

IT 9003-11-6, Ethylene oxide-propylene oxide copolymer 25322-68-3 25322-69-4, Polypropylene glycol 112344-11-3, Acrylic acid-ethylene oxide graft copolymer 146248-24-0, Ethylene-ethylene oxide-vinyl alcohol graft copolymer 273928-35-1, Crotonic acid-ethylene oxide graft copolymer 273928-36-2, Acrylic acid-crotonic acid-ethylene oxide-propylene oxide graft copolymer 276256-12-3, Acrylic acid-ethylene-ethylene oxide-propylene oxide block graft copolymer 276889-55-5, Ethylene-ethylene oxide-propylene oxide-vinyl alcohol block graft copolymer

RL: TEM (Technical or engineered material use); USES (Uses) (polyoxyalkylene containing boron complex lithium salt as ion-conducting electrolyte with improved heat resistance and corrosion inhibition)

IT 276256-12-3, Acrylic acid-ethylene-ethylene oxide-propylene oxide block graft copolymer

RL: TEM (Technical or engineered material use); USES (Uses)
(polyoxyalkylene containing boron complex lithium salt as ion-conducting
electrolyte with improved heat resistance and corrosion
inhibition)

RN 276256-12-3 HCAPLUS

CN 2-Propenoic acid, polymer with ethene, methyloxirane and oxirane, block, graft (9CI) (CA INDEX NAME)

CM 1

CRN 79-10-7 CMF C3 H4 O2 7

О || но- с- сн== сн₂

CM 2

CRN 75-56-9 CMF C3 H6 O

```
СН3
```

CRN 75-21-8 CMF C2 H4 O

 $/^{\circ}$

CM 4

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

```
L31 ANSWER 12 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN
```

AN 1999:359775 HCAPLUS

DN 131:7534

TI A proton exchange membrane fuel cell power system

IN Fuglevand, William A.; Bayyuk, Shiblihanna I.; Lloyd, Greg A.; Devries, Peter D.; Lott, David R.; Scartozzi, John P.; Somers, Gregory M.; Stokes, Ronald G.

PA Avista Labs, USA

SO PCT Int. Appl., 145 pp.

CODEN: PIXXD2

DT Patent

LA English

FAN.CNT 5

L MIN.	CMI	3																
	PAT	CENT :	NO.			KIN	D	DATE			APPL	ICAT	ION 1	NO.		D	ATE	
		-	- -				-									_		
ΡI	WO	9927	599			A1		1999	0603	1	WO 1	998-1	US21'	769		1:	9981	015 <
		W:	AL,	AM,	ΑT,	AU,	ΑZ,	BA,	BB,	BG,	BR,	BY,	CA,	CH,	CN,	CU,	CZ,	DE,
			DK,	EE,	ES,	FI,	GB,	GD,	GE,	GH,	GM,	HR,	HU,	ID,	IL,	IS,	JP,	KE,
								LK,							-	-	-	
			MX,	NO,	NZ,	PL,	PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	TJ,	TM,	TR,
			TT,	UA,	UG,	UZ,	VN,	YU,	ZW	·	•		·	•	•	•	•	•
		RW:	GH,	GM,	ΚE,	LS,	MW,	SD,	SZ,	UG,	ZW,	AT,	BE,	CH,	CY,	DE,	DK,	ES,
			FI,	FR,	GB,	GR,	ΙE,	IT,	LU,	MC,	NL,	PT,	SE,	BF,	ВJ,	CF,	CG,	CI,
		•	CM,	GA,	GN,	GW,	ML,	MR,	NE,	SN,	TD,	TG						
	US	6030	718			Α		2000	0229	1	JS 1:	997-	9798	53		19	9971	120 <
	CA	2300	846			AΑ		1999	0603	(CA 1:	998-	23008	846		19	9981	015 <
	ΑU	9910	889			A1		1999	0615	7	AU 1:	999-	1088	9		19	9981	015 <
	ΑU	7419	75			B2		2001	1213									
	BR	9814	617			Α		2000	1003	I	3R 1	998-	1461	7		19	9981	015 <
	EΡ	1040	529			A1		2000	1004]	EP 1:	998-	95354	46		19	9981	015 <
		R:	AT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT,	LI.	LU,	NL,	SE,	MC,	PT,

INDEX NAME)

```
IE, FI
     JP 2001524740
                          T2
                                            JP 2000-522640
                                20011204
                                                                    19981015 <--
     JP 3744794
                          B2
                                20060215
     US 6218035
                          B1
                                20010417
                                            US 1999-470321
                                                                    19991221 <--
                                            JP 2005-1539
     JP 2005135926
                          A2
                                20050526
                                                                    20050106 <--
                                            JP 2005-1518
     JP 2005142167
                          A2
                                20050602
                                                                    20050106 <--
PRAI US 1997-979853
                          Α
                                19971120 <--
     JP 2000-522640
                          A3
                                19981015
                                          <--
     WO 1998-US21769
                          W
                                19981015
                                          <--
AB
     A proton exchange membrane fuel cell power system (for producing elec.
     power) includes a plurality of discrete fuel cell modules having at least
     two membrane electrode diffusion assemblies, each of the membrane
     electrode diffusion assemblies having opposite anode and cathode sides; a
     pair of current collectors individually disposed in juxtaposed ohmic elec.
     contact with opposite sides of the membrane electrode diffusion
     assemblies; and individual force application assemblies applying a given
     force to the pair of current collectors and the individual membrane
     electrode diffusion assemblies. The proton exchange fuel cell power
     system also includes an enclosure mounting a plurality of subracks which
     receive the discrete fuel cell modules. Addnl., a control system is
     disclosed which optimizes the performance parameters of the discrete
     proton exchange fuel cell modules.
IC
     ICM H01M008-10
     ICS H01M008-24
CC
     52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
     Section cross-reference(s): 38
IT
     Fuel cell electrolytes
     Fuel cells
        (proton exchange membrane fuel cell power system)
     225644-20-2, 2-Propenoic acid, 2-methyl-, 3-sulfopropyl
TT
     ester-polypropylene glycol monomethacrylate-2-Propenoic acid, 2-methyl-,
     2-hydroxypropyl ester-2-Propenoic acid, 2-methyl-, 2-hydroxy-1,3-
     propanediyl ester-1,2-Dimethoxyethane-ethylene graft copolymer
     225644-21-3, 3-Sulfopropyl methacrylate-polypropylene glycol
                                  225644-22-4, 3-Sulfopropyl
     monomethacrylate copolymer
     methacrylate-polyethylene glycol monomethacrylate copolymer
                                                                    225644-63-3,
     3-Sulfopropyl methacrylate-hydroxypropyl methacrylate copolymer
     225644-64-4, 3-Allyloxy-2-hydroxy-1-propanesulfonic
     acid-polypropylene glycol monomethacrylate-hydroxypropyl
     methacrylate-diethylene glycol monomethacrylate-ethylene graft copolymer
     225644-65-5 225644-66-6
     RL: DEV (Device component use); USES (Uses)
        (proton exchange membrane fuel cell power system)
TΤ
     225644-20-2, 2-Propenoic acid, 2-methyl-, 3-sulfopropyl
     ester-polypropylene glycol monomethacrylate-2-Propenoic acid, 2-methyl-,
     2-hydroxypropyl ester-2-Propenoic acid, 2-methyl-, 2-hydroxy-1,3-
     propanediyl ester-1,2-Dimethoxyethane-ethylene graft copolymer
     225644-64-4, 3-Allyloxy-2-hydroxy-1-propanesulfonic
     acid-polypropylene glycol monomethacrylate-hydroxypropyl
     methacrylate-diethylene glycol monomethacrylate-ethylene graft copolymer
     225644-65-5 225644-66-6
     RL: DEV (Device component use); USES (Uses)
        (proton exchange membrane fuel cell power system)
RN
     225644-20-2 HCAPLUS
     2-Propenoic acid, 2-methyl-, 2-hydroxy-1,3-propanediyl ester, polymer with
CN
     1,2-dimethoxyethane, ethene, 2-hydroxypropyl 2-methyl-2-propenoate,
     \alpha-(2-methyl-1-oxo-2-propenyl)-\omega-hydroxypoly[oxy(methyl-1,2-
     ethanediyl)] and 3-sulfopropyl 2-methyl-2-propenoate, graft (9CI)
```

CRN 39420-45-6

CMF (C3 H6 O)n C4 H6 O2

CCI IDS, PMS

$$H_2C$$
 O \parallel \parallel \parallel OH $Me-C-C$ $C-C$ $0-(C_3H_6)$ $-$ OH

CM 2

CRN 7582-21-0 CMF C7 H12 O5 S

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ || & || \\ \text{HO}_3 \text{S-} & (\text{CH}_2)_3 - \text{O-} \text{C-} \text{C-} \text{Me} \end{array}$$

CM 3

CRN 1830-78-0 CMF C11 H16 O5

CM 4

CRN 923-26-2 CMF C7 H12 O3

CM 5

CRN 110-71-4 CMF C4 H10 O2

 ${\tt MeO-CH_2-CH_2-OMe}$

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 225644-64-4 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(2-hydroxyethoxy)ethyl ester, polymer with ethene, 2-hydroxy-3-(2-propenyloxy)-1-propanesulfonic acid, $\alpha\text{-}(2\text{-methyl-1-oxo-2-propenyl})-\omega\text{-hydroxypoly[oxy(methyl-1,2-ethanediyl)]}$ and 1,2-propanediol mono(2-methyl-2-propenoate), graft (9CI) (CA INDEX NAME)

CM 1

CRN 94928-31-1 CMF C6 H12 O5 S

$$\begin{array}{c} \text{OH} \\ | \\ \cdot \text{ HO}_3\text{S-CH}_2\text{-CH-CH}_2\text{-O-CH}_2\text{-CH-CH}_2 \end{array}$$

CM 2

CRN 39420-45-6 CMF (C3 H6 O)n C4 H6 O2 CCI IDS, PMS

$$H_2C$$
 O $\parallel \parallel \parallel$ \square OH $Me-C-C$ \square OH OH_6

CM 3

CRN 2351-43-1 CMF C8 H14 O4

CM 4

CRN 74-85-1 CMF C2 H4 $H_2C = CH_2$

CM 5

CRN 27813-02-1 CMF C7 H12 O3 CCI IDS

CM 6

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2 \text{H} \end{array}$$

CM 7

CRN 57-55-6 CMF C3 H8 O2

RN 225644-65-5 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, diester with 1,2,3-propanetriol, polymer with
1,1'-[1,2-ethanediylbis(oxy)]bis[ethene], ethene, 2-hydroxy-3-(2propenyloxy)-1-propanesulfonic acid, α-(2-methyl-1-oxo-2-propenyl)ω-hydroxypoly(oxy-1,2-ethanediyl) and 1,2-propanediol
mono(2-methyl-2-propenoate), graft (9CI) (CA INDEX NAME)

CM 1

CRN 94928-31-1 CMF C6 H12 O5 S

$$\begin{array}{c} \text{OH} \\ | \\ \text{HO}_{3}\text{S-CH}_{2}\text{-CH-CH}_{2}\text{-O-CH}_{2}\text{-CH-CH}_{2} \end{array}$$

CM 2

CRN 25736-86-1

CMF (C2 H4 O)n C4 H6 O2

CCI PMS

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-} & \text{C-} & \text{C-} & \text{C-} & \text{CH}_2 - \text{CH}_2 - \text{CH}_2 \\ \end{array} \begin{array}{c} \text{OH} \\ \end{array}$$

CRN 764-78-3 CMF C6 H10 O2

$$H_2C = CH - O - CH_2 - CH_2 - O - CH = CH_2$$

CM 4

CRN 74-85-1 CMF C2 H4

$$H_2C = CH_2$$

CM 5

CRN 28497-59-8 CMF C11 H16 O5 CCI IDS

CM 6

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2 \text{H} \end{array}$$

CM 7

CRN 56-81-5 CMF C3 H8 O3

$$\begin{array}{c} \text{OH} \\ | \\ \text{HO- CH}_2\text{-- CH-- CH}_2\text{-- OH} \end{array}$$

CM 8

27813-02-1 CRN C7 H12 O3 CMF

CCI IDS

> CM 9

CRN 79-41-4 CMF C4 H6 O2

CH₂ Me- C- CO2H

> CM10

CRN 57-55-6 CMF C3 H8 O2

OH $H_3C-CH-CH_2-OH$

RN 225644-66-6 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, monoester with 1,2-propanediol, polymer with bis(2-propenyloxy)acetic acid, 1,1'-[1,2-ethanediylbis(oxy)]bis[ethene], ethene, 2-hydroxy-3-(2-propenyloxy)-1-propanesulfonic acid and α -(2-methyl-1-oxo-2-propenyl)- ω -hydroxypoly(oxy-1,2ethanediyl), graft (9CI) (CA INDEX NAME)

CM 1

CRN 161823-92-3 CMF C8 H12 O4

 $O-CH_2-CH=-CH_2$ $H_2C = CH - CH_2 - O - CH - CO_2H$

> CM 2

CRN 94928-31-1 CMF C6 H12 O5 S

OH $HO_3S-CH_2-CH-CH_2-O-CH_2-CH=-CH_2$

> CM 3

CRN 25736-86-1

CMF (C2 H4 O)n C4 H6 O2

CCI PMS

CM 4

CRN 764-78-3 CMF C6 H10 O2

$$H_2C = CH - O - CH_2 - CH_2 - O - CH = CH_2$$

CM 5

CRN 74-85-1 CMF C2 H4

$$H_2C = CH_2$$

CM 6

CRN 27813-02-1

CMF C7 H12 O3

CCI IDS

CM 7

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2 \text{H} \end{array}$$

CM 8

CRN 57-55-6 CMF C3 H8 O2

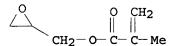
```
RE.CNT 13
              THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS RECORD
              ALL CITATIONS AVAILABLE IN THE RE FORMAT
L31 ANSWER 13 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN
AN
     1996:29908 HCAPLUS
DN
     124:69824
     Composite electrolyte for primary and secondary batteries,
TТ
     electrochromic devices, and sensors
     Okuyama, Kazuo; Suzuki, Yoshio; Ai, Hideo
IN
     Asahi Chemical Ind, Japan
PA
     Jpn. Kokai Tokkyo Koho, 8 pp.
SO
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
FAN.CNT 1
     PATENT NO.
                        KIND
                                DATE
                                           APPLICATION NO.
                                                                  DATE
                        ----
                                -----
                                           -----
                                                                   _____
     JP 07296634
PΤ
                         A2
                                19951110
                                          JP 1994-84457
                                                                   19940422 <--
PRAI JP 1994-84457
                                19940422 <--
AΒ
     The title composite electrolyte comprises (1) a core material
     made of polyolefin having proton-conductive functional group on the whole
     area and exchange capacity of 1-15 meq./g and (2) an ion-exchange resin
     having exchange capacity of 0.9-15 meq./q. The ion-exchange resin is
     fluoro ion-exchange resin having exchange capacity of 0.9-2 meg./q. It is
     required that the polyolefin has a sulfonic acid group as a
     proton-conductive functional group. A high-performance
     electrolyte film having small resistance can be provided.
IC
     ICM H01B001-06
     ICS B01J047-12; C08F010-00; C08J005-22; H01M006-18; H01M008-02
CC
     72-2 (Electrochemistry)
     Section cross-reference(s): 52, 74, 79
ST
     composite electrolyte polyolefin ion exchange resin
IT
    Batteries, primary
    Batteries, secondary
     Sensors
        (composite electrolytes for)
ΙT
     Sulfonic acids, uses
    RL: NUU (Other use, unclassified); USES (Uses)
        (perfluorocarbonsulfonic acid membranes; for preparing composite
       electrolyte)
TT
    Optical imaging devices
        (electrochromic, composite electrolytes for)
TT
    Fluoropolymers
    RL: NUU (Other use, unclassified); USES (Uses)
        (sulfo-containing, membrane; for preparing composite electrolyte)
     119433-93-1, Ethylene-glycidyl methacrylate graft copolymer
TT
     120668-45-3, Hipore 2100 148602-36-2, Acrylic acid-ethylene-sodium
     styrenesulfonate graft copolymer 172323-09-0, Acrylic acid-sodium
     styrenesulfonate-tetrafluoroethylene graft copolymer
    RL: NUU (Other use, unclassified); USES (Uses)
```

RN 119433-93-1 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester, polymer with ethene, graft (9CI) (CA INDEX NAME)

CM 1

CRN 106-91-2 CMF C7 H10 O3



CM 2

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

L31 ANSWER 14 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1995:934785 HCAPLUS

DN 124:12268

TI Polypropylene separator grafted with hydrophilic monomers for lithium batteries

AU Gineste, Jean Luc; Pourcelly, Gerald

CS Laboratoire de Materiaux et Procedes Membranaires, UMR 9987 CNRS, BP 5051, Montpellier, 34033, Fr.

SO Journal of Membrane Science (1995), 107(1-2), 155-64 CODEN: JMESDO; ISSN: 0376-7388

PB Elsevier

DT Journal

LA English

AB Acrylic acid and diethylene glycol dimethacrylate were grafted onto 50 µm polypropylene films. The physicochem. properties of the polymer films obtained were studied vs. the characteristics of grafting. The influence of temperature and monomer content on grafting kinetics is pointed out. Cycling performances of secondary lithium batteries including these grafted films as separators are also presented.

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
 Section cross-reference(s): 38

IT Electric resistance

(of grafted polypropylene battery separator as function of **electrolyte** composition)

IT Electric conductivity and conduction

(of lithium hexafluoroarsenate **electrolyte** containing propylene carbonate, ethylene carbonate, and dimethoxyethane)

IT 29935-35-1, Lithium hexafluoroarsenate

RL: DEV (Device component use); USES (Uses)

(electrolyte; performance of lithium batteries with polypropylene separator grafted with acrylic acid and diethylene glycol dimethacrylate)

IT 96-49-1, Ethylene carbonate 108-32-7, Propylene carbonate 110-71-4 RL: DEV (Device component use); USES (Uses)

(lithium hexafluoroarsenate **electrolyte** containing; performance of lithium batteries with polypropylene separator grafted with acrylic acid and diethylene glycol dimethacrylate)

IT 171370-46-0P

RL: DEV (Device component use); PEP (Physical, engineering or chemical

process); PRP (Properties); SPN (Synthetic preparation); PREP
(Preparation); PROC (Process); USES (Uses)

(polypropylene separator grafted with acrylic acid and diethylene glycol dimethacrylate for lithium batteries)

IT 171370-46-0P

RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process); USES (Uses)

(polypropylene separator grafted with acrylic acid and diethylene glycol dimethacrylate for lithium batteries)

RN 171370-46-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, oxydi-2,1-ethanediyl ester, polymer with 1-propene and 2-propenoic acid, graft (9CI) (CA INDEX NAME)

CM 1

CRN 2358-84-1 CMF C12 H18 O5

CM 2

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

CM 3

CRN 79-10-7 CMF C3 H4 O2

L31 ANSWER 15 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1994:558424 HCAPLUS

DN 121:158424

TI Preparation of modified olefin polymer particles

IN Sato, Hiroyuki

PA Mitsubishi Petrochemical Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 10 pp. CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO.

KIND DATE

APPLICATION NO.

DATE

 $^{\rm H_2C}_{||}$ $^{\rm O}_{||}$ $^{\rm H_2}_{||}$ $^{\rm Me-}$ $^{\rm C-}$ $^{\rm C-}$ $^{\rm O-}$ $^{\rm CH_2-}$ $^{\rm CH_2-}$ $^{\rm OH}$

CM 2

CRN 74-85-1 CMF C2 H4 $H_2C = CH_2$

RN 116945-18-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester, polymer with ethenylbenzene and 1-propene, graft (9CI) (CA INDEX NAME)

CM 1

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

CM 2

CRN 106-91-2 CMF C7 H10 O3

CM 3

CRN 100-42-5 CMF C8 H8

H2C=CH-Ph

RN 119028-94-3 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-hydroxyethyl ester, polymer with 1-propene, graft (9CI) (CA INDEX NAME)

CM 1

CRN 868-77-9 CMF C6 H10 O3

$$^{\rm H_2C}$$
 O $^{\parallel}$ $^{\parallel}$ $^{\parallel}$ Me- C- C- O- CH₂- CH₂- OH

CM 2

CRN 115-07-1 CMF C3 H6 $H_3C-CH=CH_2$

RN 119433-93-1 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester, polymer with ethene, graft (9CI) (CA INDEX NAME)

CM 1

CRN 106-91-2 CMF C7 H10 O3

$$\begin{array}{c|c} O & CH_2 \\ & \parallel & \parallel \\ CH_2-O-C-C-Me \end{array}$$

CM 2

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 148527-10-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-hydroxyethyl ester, polymer with ethene and ethenylbenzene, graft (9CI) (CA INDEX NAME)

CM 1

CRN 868-77-9 CMF C6 H10 O3

$$^{\rm H_2C}$$
 о \parallel \parallel \parallel $^{\rm Me-C-C-O-CH_2-CH_2-OH}$

CM 2

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

CM 3

CRN 74-85-1 CMF C2 H4

```
H_2C = CH_2
```

```
L31 ANSWER 16 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN
     1993:542974 HCAPLUS
AN
DN
     119:142974
TI
     Secondary batteries with improved separators
IN
     Kubota, Tadahiko
     Fuji Photo Film Co Ltd, Japan
PA
     Jpn. Kokai Tokkyo Koho, 9 pp.
so
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
FAN.CNT 1
     PATENT NO.
                                           APPLICATION NO.
                         KIND
                                DATE
                                                                   DATE
                         ____
                                _____
                                            -----
                                                                   _____
                                            JP 1991-265475
PΙ
     JP 05082114
                         A2
                                19930402
                                                                   19910918 <--
     US 5270137
                         Α
                                19931214
                                           US 1992-944986
                                                                   19920915 <--
PRAI JP 1991-265475
                         Α
                                19910918 <--
     The batteries use polymer separators, obtained by treating a porous
     polymer substrate with plasma and grafting monomers on the substrate.
     monomers may contain groups soluble in the battery electrolyte, and
     are preferably CH2:CR1R2 (R1 = H or alkyl, R2 = COX1(CH2CH2O)nR3 or
     COX1(CH2)xPh, or CN, X = O or NR4, R4 = H or alkyl, n is integer, R3 = H,
     C1-3 alkyl, aralkyl, aryl, x is integer), (2) CH2:CR5X2CR5:CH2 (R5 = H or
     alkyl, X2 = COX3CO, X3 = OX4O or O(CH2CH2O)m, X4 = alkylene, m is
     \geq1 integer), or (3) (CH2:CR6CO2CH2)3CR7 (R6 = H or alkyl, R7 =
     alkyl or CH2OR8, R8 = H or alkyl). The batteries using these separators
     have suppressed dendrite growth and long cycle life.
     ICM H01M002-16
IC .
CC
     52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
     Section cross-reference(s): 38
     137091-22-6P 149797-11-5P 149797-12-6P
     149830-04-6P 149855-93-6P 149855-94-7P
     149855-95-8P 149855-96-9P 149855-97-0P
     RL: PREP (Preparation)
        (separator, manufacture of, for batteries)
     149797-11-5P 149797-12-6P 149830-04-6P
IT
     149855-93-6P 149855-95-8P 149855-96-9P
     149855-97-0P
     RL: PREP (Preparation)
        (separator, manufacture of, for batteries)
     149797-11-5 HCAPLUS
RN
CN
     Poly(oxy-1,2-ethanediyl), \alpha-(2-methyl-1-oxo-2-propenyl)-\omega-
     phenoxy-, polymer with 1-propene, graft (9CI) (CA INDEX NAME)
     CM
          1
     CRN
         50858-63-4
          (C2 H4 O)n C10 H10 O2
     CMF
     CCI PMS
```

H₂C

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

RN 149797-12-6 HCAPLUS CN Poly(oxy-1,2-ethanediyl), α -(2-methyl-1-oxo-2-propenyl)- ω -phenoxy-, polymer with α -(1-oxo-2-propenyl)- ω -[(1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl) and 1-propene, graft (9CI) (CA INDEX NAME)

CM 1

CRN 50858-63-4

CMF (C2 H4 O)n C10 H10 O2

CCI PMS

CM 2

CRN 26570-48-9

CMF (C2 H4 O)n C6 H6 O3

CCI PMS

$$H_2C = CH - C - CH_2 - CH_2$$

CM 3

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

RN 149830-04-6 HCAPLUS CN 2-Propenoic acid, 2-methyl-, phenylmethyl ester, polymer with α -(1-oxo-2-propenyl)- ω -[(1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl) and 1-propene, graft (9CI) (CA INDEX NAME)

CM 1

26570-48-9 CRN CMF (C2 H4 O)n C6 H6 O3 CCI PMS

$$H_2C = CH - C - CH_2 - CH_2$$

CM 2

CRN 2495-37-6 CMF C11 H12 O2

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-} & \text{C-} & \text{C-} & \text{O-} & \text{CH}_2 - \text{Ph} \end{array}$$

CM 3

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

149855-93-6 HCAPLUS RN2-Propenoic acid, 2-methyl-, 2-ethyl-2-[[(2-methyl-1-oxo-2-CNpropenyl)oxy]methyl]-1,3-propanediyl ester, polymer with α -(2-methyl-1-oxo-2-propenyl)- ω -methoxypoly(oxy-1,2ethanediyl), α -(2-methyl-1-oxo-2-propenyl)- ω -[(2-methyl-1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl) and 1-propene, graft (9CI) INDEX NAME)

CM 1

CRN 26915-72-0 CMF (C2 H4 O)n C5 H8 O2 CCI PMS

$$\begin{array}{c|c} {}^{\rm H_2C} & {}^{\rm O} \\ \parallel & \parallel \\ {}^{\rm Me-} & {}^{\rm C-} & {}^{\rm C-} & {}^{\rm C-} & {}^{\rm C-} & {}^{\rm C} + {}^{\rm C} \\ \end{array}$$

CM 2

CRN 25852-47-5 CMF (C2 H4 O)n C8 H10 O3 CCI PMS

CM 3

CRN 3290-92-4 CMF C18 H26 O6

CM 4

CRN 115-07-1 CMF C3 H6

RN 149855-95-8 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), α -(1-oxo-2-propenyl)- ω -methoxy-, polymer with 1-propene, graft (9CI) (CA INDEX NAME)

CM 1

CRN 32171-39-4

CMF (C2 H4 O)n C4 H6 O2

CCI PMS

$$H_2C = CH - C - O - CH_2 - CH_2 - OMe$$

CM 2

CRN 115-07-1 CMF C3 H6 $H_3C-CH=CH_2$

RN 149855-96-9 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-ethyl-2-[[(2-methyl-1-oxo-2-propenyl)oxy]methyl]-1,3-propanediyl ester, polymer with α -(2-methyl-1-oxo-2-propenyl)- ω -[(2-methyl-1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl) and 1-propene, graft (9CI) (CA INDEX NAME)

CM 1

CRN 25852-47-5

CMF (C2 H4 O)n C8 H10 O3

CCI PMS

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-} & \text{C-} & \text{C-} & \text{C-} & \text{CH}_2 - \text{CH}_2 - \text{CH}_2 \\ \end{array} \\ \begin{array}{c|c} \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{O} & \text{C-} & \text{C-} & \text{Me} \end{array}$$

CM 2

CRN 3290-92-4 CMF C18 H26 O6

CM 3

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

RN 149855-97-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-ethyl-2-[[(2-methyl-1-oxo-2-propenyl)oxy]methyl]-1,3-propanediyl ester, polymer with α -(1-oxo-2-propenyl)- ω -[(1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl) and 1-propene, graft (9CI) (CA INDEX NAME)

CM 1

CRN 26570-48-9

CMF (C2 H4 O)n C6 H6 O3

CCI PMS

$$H_2C = CH - C - CH_2 - CH_2$$

CM 2

CRN 3290-92-4 CMF C18 H26 O6

CM 3

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

L31 ANSWER 17 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1989:39896 HCAPLUS

DN 110:39896

TI Heat-resistant polyolefin blend compositions

IN Iwashita, Toshiyuki; Mogi, Yoshihiro; Kurosawa, Hayashi; Funada, Hitoshi

PA Showa Denko K. K., Japan

SO Jpn. Kokai Tokkyo Koho, 17 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE			
PI	JP 63159420	A2	19880702	JP 1986-306364	19861224 <			
	JP 06057737	B4	19940803					
PRAI	JP 1986-306364		19861224	<				

AB The title compns. with good transparency and elec. property and useful in preparing printed board substrates, flexible printed boards, potting compns. for electrolyte capacitors, etc. are prepared from olefin-unsatd. monocarboxylic acid copolymers, saponified and demetalized olefin-unsatd. carboxylate copolymers, and/or olefin-dicarboxylic acid (anhydride or half ester) copolymers 1-99, olefin-epoxy-containing unsatd. compound copolymers 1-99, and carboxylic acid ester plasticizers 1-50 parts. A mixture of acrylic acid-ethylene copolymer 50, ethylene-glycidyl methacrylate-vinyl

acetate copolymer 50, and di-Bu phthalate (I) 10 parts were molded to give a film having gel fraction 85.4%, peeling strength (of Cu foil laminate) 3.2 kg/25 mm, and haze (after heated 36 h at 60°) 0.8%, vs. 93.6, 6.2, and 2.6, resp., for a film without I.

IC ICM C08G059-18

ICS C08G059-18; C08G059-32; C08L023-08; C08L063-00

ICA C08K005-10

C 37-6 (Plastics Manufacture and Processing)

IT 36604-80-5, Ethylene-glycidyl methacrylate-vinyl acetate copolymer
52642-93-0

RL: USES (Uses)

(blends with ethylene-acrylate copolymers and plasticizers, transparent and heat-resistant)

IT 36604-80-5, Ethylene-glycidyl methacrylate-vinyl acetate copolymer 52642-93-0

RL: USES (Uses)

(blends with ethylene-acrylate copolymers and plasticizers, transparent and heat-resistant)

RN 36604-80-5 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester, polymer with ethene and ethenyl acetate (9CI) (CA INDEX NAME)

CM 1

CRN 108-05-4 CMF C4 H6 O2

 $AcO-CH=CH_2$

CM 2

CRN 106-91-2 CMF C7 H10 O3

$$\begin{tabular}{c|c} O & CH_2 \\ \hline & \parallel & \parallel \\ CH_2-O-C-C-Me \end{tabular}$$

CM 3

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

RN 52642-93-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, methyl ester, polymer with ethene and [(2-propenyloxy)methyl]oxirane (9CI) (CA INDEX NAME)

CM 1

CRN 106-92-3

CMF C6 H10 O2

CM 2

CRN 80-62-6 CMF C5 H8 O2

$$H_2C$$
 O | | || Me- C- C- OMe

CM 3

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

L31 ANSWER 18 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1987:423847 HCAPLUS

DN 107:23847

TI Synthesis of graft polyethylene having polyelectrolyte complex structure

AU Murata, Kenichi; Hayashi, Kazuko

CS Japan

SO Osaka Kogyo Gijutsu Shikensho Kiho (1987), 38(1), 1-7

CODEN: OKGKAE; ISSN: 0472-142X

DT Journal

LA Japanese

- Preirradiated polyethylene was graft copolymd. with 2-(dimethylamino)ethyl methacrylate, and the resulting graft copolymer was treated with EtBr or HBr, and subsequently with Na p-styrenesulfonate. The p-styrenesulfonate groups in the graft copolymers were polymerized using (NH4)2S2O8 initiator in water. The stability of the resulting polyelectrolyte complexes in electrolytic solns. was investigated by immersing in 0.5 N NaBr, 0.5 N HBr, and 0.5 N and 1.0 N NaOH at 30° for 25 h. Although the complexes were stable to salt and acid, 20.apprx.25% was eluted by alkaline solution
- CC 35-8 (Chemistry of Synthetic High Polymers)
- IT 108811-95-6P 108811-96-7P 108811-97-8P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and polymerization of)

IT 107227-29-2P 108811-91-2P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and quaternization of)

IT 108811-92-3P 108811-93-4P 108811-94-5P

```
WEINER 10/828468 09/12/2006 Page 68
```

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and reaction of, with sodium styrenesulfonate)

IT 108811-88-7P 108811-89-8P 108811-90-1P **108828-62-2P** RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

IT 108811-97-8P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and polymerization of)

RN 108811-97-8 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(dimethylamino)ethyl ester, polymer with ethene and 2-hydroxyethyl 2-methyl-2-propenoate, graft, compd. with bromoethane and sodium 4-ethenylbenzenesulfonate (9CI) (CA INDEX NAME)

CM 1

CRN 2695-37-6 CMF C8 H8 O3 S . Na

Na

CM 2

CRN 108811-94-5

CMF (C8 H15 N O2 . C6 H10 O3 . C2 H4)x . x C2 H5 Br

CM 3

CRN 74-96-4 CMF C2 H5 Br

Br-CH2-CH3

CM 4

CRN 108811-91-2

CMF (C8 H15 N O2 . C6 H10 O3 . C2 H4) \times

CCI PMS

CM 5

CRN 2867-47-2 CMF C8 H15 N O2

$$\begin{array}{c} \text{O} \quad \text{CH}_2 \\ \parallel \quad \parallel \\ \text{Me}_2 \text{N-CH}_2 - \text{CH}_2 - \text{O-C-C-Me} \end{array}$$

CRN 868-77-9 CMF C6 H10 O3

CM 7

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

IT 108811-91-2P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and quaternization of)

RN 108811-91-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(dimethylamino)ethyl ester, polymer with ethene and 2-hydroxyethyl 2-methyl-2-propenoate, graft (9CI) (CA INDEX NAME)

CM 1

CRN 2867-47-2 CMF C8 H15 N O2

$$\begin{array}{c} \text{O} \quad \text{CH}_2 \\ \parallel \quad \parallel \\ \text{Me}_2 \text{N-CH}_2 - \text{CH}_2 - \text{O-C-C-Me} \end{array}$$

CM 2

CRN 868-77-9 CMF C6 H10 O3

WEINER 10/828468 · 09/12/2006 Page 70

CM 3

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

IT 108811-94-5P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and reaction of, with sodium styrenesulfonate)

RN 108811-94-5 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(dimethylamino)ethyl ester, polymer with ethene and 2-hydroxyethyl 2-methyl-2-propenoate, graft, compd. with bromoethane (9CI) (CA INDEX NAME)

CM 1

CRN 74-96-4 CMF C2 H5 Br

 $\operatorname{Br-CH}_2-\operatorname{CH}_3$

CM 2

CRN 108811-91-2

CMF (C8 H15 N O2 . C6 H10 O3 . C2 H4)x

CCI PMS

CM 3

CRN 2867-47-2 CMF C8 H15 N O2

$$\begin{array}{c} {\rm O} \quad {\rm CH_2} \\ || \quad || \\ {\rm Me_2N-CH_2-CH_2-O-C-C-Me} \end{array}$$

CM 4

CRN 868-77-9 CMF C6 H10 O3

$$^{\rm H_2C}_{\parallel}$$
 0 $^{\rm Me-C-C-O-CH_2-CH_2-OH}_{\parallel}$

CM 5

WEINER 10/828468 - 09/12/2006

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

IT 108828-62-2P

RL: SPN (Synthetic preparation); PREP (Preparation)
 (preparation of)

RN 108828-62-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(dimethylamino)ethyl ester, polymer with ethene and 2-hydroxyethyl 2-methyl-2-propenoate, graft, compd. with bromoethane and sodium 4-ethenylbenzenesulfonate homopolymer (9CI) (CA INDEX NAME)

Page 71

CM 1

CRN 108811-94-5

CMF (C8 H15 N O2 . C6 H10 O3 . C2 H4)x . x C2 H5 Br

CM 2

CRN 74-96-4 CMF C2 H5 Br

Br-CH2-CH3

CM 3

CRN 108811-91-2

CMF (C8 H15 N O2 . C6 H10 O3 . C2 H4) \mathbf{x}

CCI PMS

CM 4

CRN 2867-47-2 CMF C8 H15 N O2

CM 5

CRN 868-77-9 CMF C6 H10 O3

 $\begin{array}{c|c} ^{\rm H_2C} & {\rm O} \\ & \parallel & \parallel \\ ^{\rm Me-} & {\rm C-C-O-CH_2-CH_2-OH} \end{array}$

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

CM 7

CRN 25704-18-1

CMF (C8 H8 O3 S . Na)x

CCI PMS

CM 8

CRN 2695-37-6

CMF C8 H8 O3 S . Na

Na

L31 ANSWER 19 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1983:217055 HCAPLUS

DN 98:217055

TI Synthetic thickener for printing pastes

AU Terteryan, R. A.; Ivanov, V. I.; Khrapov, V. S.; Repina, E. V.; Senakhov, A. V.; Lomakina, T. N.

CS VNIINP, Moscow, USSR

SO Lakokrasochnye Materialy i Ikh Primenenie (1983), (2), 10-13 CODEN: LAMAAD; ISSN: 0023-737X

DT Journal

LA Russian

AB The effect is shown of mol. weight, crosslinking agent type, and conditions of preparation of ethylene-maleic anhydride copolymer (I) [9006-26-2] on its rheol. and printing properties as a thickner. The apparent viscosity of I solns. linearly increased as logarithmic function of its mol. weight, and was optimal at the mol. weight .apprx.106. I with this mol. weight was obtained by polymerization at .apprx.150 MPa. At lower pressures,

suitable thickness were prepared by polymerization in the presence of **crosslinking** agents. divinylbenzene-ethylene-maleic anhydride copolymer [85885-43-4] Had the lowest viscosity and thixotropic recover. Acceptable thixotropic recovery (75-85%) and viscosity (7.2-9.9 Pa-s at shear rate gradient 48.6s) were exhibited by 3-6% aqueous solns. of thickeners containing 2.4-5 mol. % allyl methacrylate (II) or ethylene glycol dimethacrylate (III) units. In most cases, the viscosity of thickeners

increased with increasing content of **crosslinking** agents, whereas thixotropic recovery decreased. The thickeners containing II units were recommended for the pastes with low ionic force, while those containing III units for pastes with reactive dyes, due to higher resistance to low.-mol.-weight **electrolytes** of the latter as compared to that of the formed.

CC 40-6 (Textiles)

ST thickener ethylene maleic anhydride copolymer; printing paste thickener crosslinking agent; viscosity printing thickener mol wt; thixotropy printing thickener crosslinking agent; rheol printing thickener crosslinking agent; polymn printing thickener crosslinking agent

IT Crosslinking agents

(for ethylene-maleic anhydride copolymer thickeners, for textile printing)

IT 9006-26-2 37309-45-8 77036-17-0 **85874-70-0**

85874-71-1

RL: USES (Uses)

(thickeners, for textile printing pastes, rheol. of)

IT 85874-70-0 85874-71-1

RL: USES (Uses)

(thickeners, for textile printing pastes, rheol. of)

RN 85874-70-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 1,2-ethanediyl ester, polymer with ethene and 2,5-furandione (9CI) (CA INDEX NAME)

CM 1

CRN 108-31-6 CMF C4 H2 O3

CM 2

CRN 97-90-5 CMF C10 H14 O4

CM 3

CRN 74-85-1 CMF C2 H4

H2C= CH2

RN 85874-71-1 HCAPLUS

WEINER 10/828468 09/12/2006 Page 74

CN 2-Propenoic acid, 2-methyl-, oxybis(2,1-ethanediyloxy-2,1-ethanediyl) ester, polymer with ethene and 2,5-furandione (9CI) (CA INDEX NAME)

CM 1

CRN 109-17-1 CMF C16 H26 O7

PAGE 1-B

--- Me

CM 2

CRN 108-31-6 CMF C4 H2 O3

0 0

CM 3

CRN 74-85-1 CMF C2 H4

 $H_2C = CH_2$

L31 ANSWER 20 OF 20 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1981:489975 HCAPLUS

DN 95:89975

TI Electrolytic capacitor

PA Matsushita Electric Industrial Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 4 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PRAI JP 1979-125747 A 19790928 <--

AB A corrosion-resistant electrolytic capacitor is obtained by placing a separator soaked with an electrolytic solution between electrode foils, rolling to form an interior element, attaching pull-out leads to the interior element, sealing with a packaging film to form a capacitor element, placing the capacitor element in an exterior case having terminals, and connecting the leads and terminals. A laminate from metal and resin (e.g., polyolefin, ionomer resin) films may be used for packaging.

IC H01G009-08

CC 76-3 (Electric Phenomena)

ST **electrolytic** capacitor ionomer laminate; metal laminate **electrolytic** capacitor; polyolefin laminate **electrolytic** capacitor; olefin polymer laminate **electrolytic** capacitor

IT Alkenes, polymers

Ionomers

Metals, uses and miscellaneous

Polyesters, uses and miscellaneous

RL: USES (Uses)

(electrolytic capacitor packaging by laminated films from)

IT Electric capacitors

(electrolytic, laminate packaging for)

TT 7429-90-5, uses and miscellaneous 9002-88-4 9010-77-9 9078-96-0 36604-80-5 78690-54-7

RL: USES (Uses)

(electrolytic capacitor packaging by laminated films from)

IT 36604-80-5

RL: USES (Uses)

(electrolytic capacitor packaging by laminated films from)

RN 36604-80-5 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester, polymer with ethene and ethenyl acetate (9CI) (CA INDEX NAME)

CM 1

CRN 108-05-4 CMF C4 H6 O2

Aco-CH-CH2

CM 2

CRN 106-91-2 CMF C7 H10 O3

 $\begin{tabular}{c|c} O & O & CH_2 \\ & \parallel & \parallel \\ CH_2-O-C-C-Me \end{tabular}$

CM 3

CRN 74-85-1 CMF C2 H4 WEINER 10/828468 09/12/2006 Page 76

 $H_2C = CH_2$

= >

Kunio Maruyama et al., S.N. 10/828,468 Page 2

functional groups (that is, vinyl group, epoxy group, amino group, amide group, imide group, hydroxyl group, methylol group, carboxyl group and isocyanate group),

- (ii) select one of the non-crosslinked polymer in claim 1, including identifying whether it comprises an ethylene unit, a propylene unit or both, and
- (iii) select one of the polyalkylene glycol in claim 3 (that is, polyethylene glycol, polypropylene glycol or polyethylene/propylene glycol).

In response to the species election requirement, Applicants hereby elect, with traverse, to prosecute the species comprising (i) a crosslinkable monomer having a methylol group and an isocyanate group, (ii) a non-crosslinked polymer having an ethylene unit and a propylene unit, and (iii) polyethylene glycol.

018

As discussed previously of record, Applicants, however, respectfully request that the Examiner reconsider and withdraw the requirement.

Applicants maintain that the various species are not independent. In addition, Applicants further maintain that it would not be a serious burden on the Examiner if restriction is not required, because a search of the prior art for one species would likely identify art for other species. Accordingly, the Examiner should examine all of the species covered by the claims on the merits.

Accordingly, in view of the preceding remarks, Applicants respectfully request that the Examiner reconsider and withdraw the restriction requirement.

No fee is deemed necessary in connection with the filing of this Communication. However, if any fee is required, authorization is